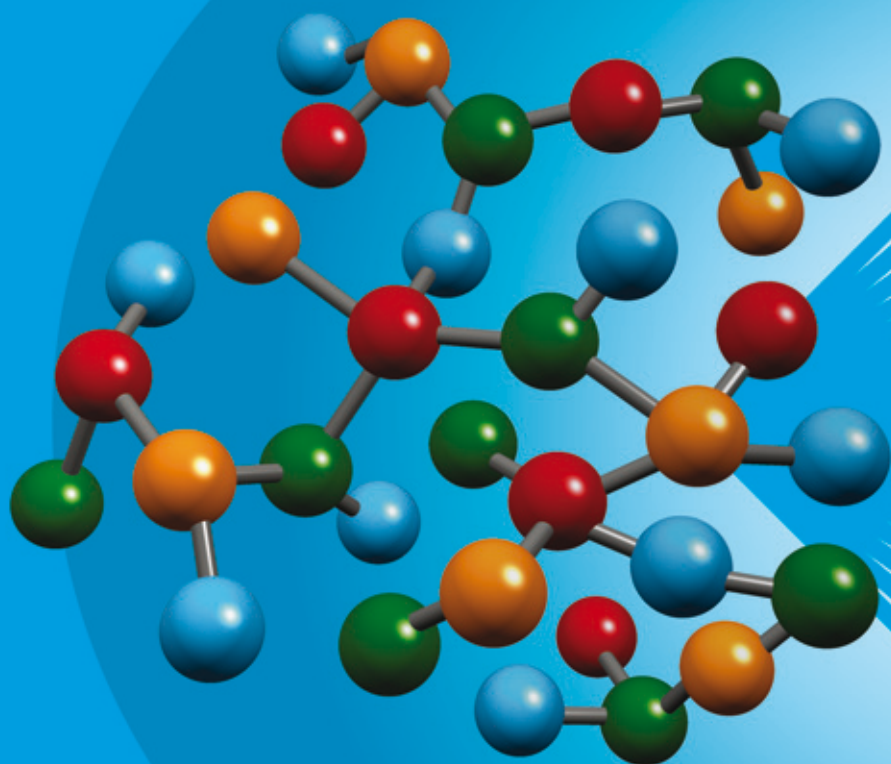


FREE ACADEMIC VERSION

10

USER MANUAL



PARASURETM

Impressum

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PROGRAM HISTORY

Release Date	Version	Platforms
1 st July 2005	ParaSurf 05 TM initial release (Revision A1)	32-bit Windows 32-bit Linux Irix
1 st January 2006	ParaSurf 05 TM Revision B1 (customer-feedback release)	
1 st July 2006	ParaSurf 06 TM Revision A1	32-bit Windows 32-bit Linux 64-bit Linux Irix
1 st July 2007	ParaSurf 07 TM Revision A1	
1 st July 2008	ParaSurf 08 TM Revision A1	32-bit Windows 64-bit Windows 32-bit Linux 64-bit Linux
22 nd August 2008	ParaSurf 08 TM Revision A2 (minor bug fix release)	
16 th December 2008	ParaSurf 08 TM Revision A3 (minor bug fix release)	
1 st July 2009	ParaSurf 09 TM Revision A1	
1 st September 2009	New Vhamil.par file including PM6 and first-row transition metals in AM1*	
1 st February 2010	ParaSurf 09 TM Revision B1 (additional atom-centred descriptors)	
1 st July 2010	ParaSurf 10 TM Revision A1	
1 st July 2011	ParaSurf 10 TM Academic Version	

1 INTRODUCTION

ParaSurf[™] is a program to generate isodensity or solvent-excluded surfaces from the results of semiempirical molecular orbital calculations, either from VAMP [1] or a public-domain version of MOPAC modified and made available by Cepos InSilico. [2] The surface may be generated by shrink-wrap [3] or marching-cube [4] algorithms and the former may be fit to a spherical harmonic series. [5] The principles of these two techniques are explained below, but for comparison Figure 1 shows default isodensity surfaces calculated by ParaSurf[™] for a tetracycline derivative. The surfaces are color-coded according to the electrostatic potential at the surface.

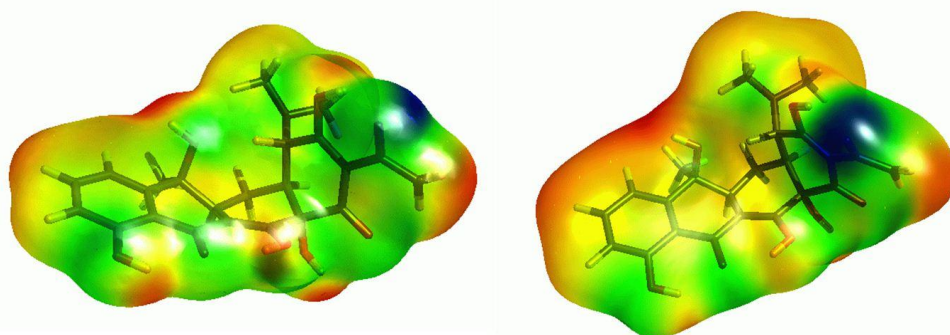


Figure 1 Marching-cube (left) and shrink-wrap (right, fitted to a spherical-harmonic approximation) isodensity surfaces calculated with ParaSurf[™] using the default settings

Four local properties, the molecular electrostatic potential (MEP), [6] the local ionization energy (IE_L), [7] the local electron affinity (EA_L), [8] and the local polarisability (α_L) [8] are calculated at the points on the surface. Two further properties, the local hardness (η_L), [8] and the local electronegativity (χ_L) [8] can be derived from IE_L and EA_L .

The local properties can be used to generate a standard set of 81 descriptors [9] appropriate for quantitative structure-property relationships (QSPRs) for determining physical properties.

ParaSurf[™] can also generate local enthalpies and free energies of solvation [10] and integrate them over the entire molecular surface to give the enthalpy or free energy of solvation. ParaSurf[™] can read so-called *Surface-Integral Model* (SIM) files that allow it to calculate properties such as, for instance, the enthalpy and free energy of hydration and the free energies of solvation in *n*-octanol and chloroform. The surface-integral models are expressed as summations of local solvation energies over the molecular surface. These local solvation energies can be written to the ParaSurf[™] surface file.

ParaSurf[™] is the first program to emerge from the ParaShift collaboration between researchers at the Universities of Erlangen, Portsmouth, Southampton, Oxford and Aberdeen. It is intended to provide the molecular surfaces for small molecules (i.e. non-proteins) for subsequent quantitative structure-activity relationship (QSAR), QSPR, high-throughput virtual screening (HTVS), docking and scoring, pattern-recognition and simulation software that will be developed in the ParaShift project.



1.1 Changes relative to ParaSurf 09™

ParaSurf 10™ has been enhanced relative to its predecessor in order to provide improved flexibility and a more comprehensive range of descriptors and features. The changes are outlined below:

1.1.1 Local electron affinity for AM1* and other Hamiltonians with *d*-orbitals as polarisation functions

Calculating the local electron affinity with AM1* led to spurious results with ParaSurf 09™ because the *d*-polarisation functions dominated the summation. A new technique [11] has been introduced to fix this problem in ParaSurf 10™. A new command-line option requests that the local electron affinity be calculated exactly as in ParaSurf 09™ to ensure continuity.

1.1.2 The PM6 Hamiltonian

ParaSurf 10™ can be used with PM6. [12] This was also the case with ParaSurf 09 if the post-release Vhamil.par file were used. PM6 is available for 70 elements.

1.1.3 Second generation surface-integral models; local hydrophobicity*

1.1.4 Molecular fragments*

1.1.5 Atom-centred descriptors*

1.2 Isodensity surfaces

Isodensity surfaces [13] are defined as the surfaces around a molecule at which the electron density has a constant value. Usually this value is chosen to approximate the van der Waals' shape of the molecule. ParaSurf™ allows values of the isodensity level down to $0.00001 \text{ e}^- \text{ \AA}^{-3}$. Lower values than this may result in failures of the surface algorithms for very diffuse surfaces.

1.3 Solvent-excluded surfaces

The solvent-excluded surface is obtained by rolling a spherical solvent molecule of radius r_{solv} over the surface of the molecule as shown in Figure 2. The surface of the solvent molecule defines the molecular surface, so that the yellow volume in Figure 2 becomes part of the molecule.

* Only available in the full version.

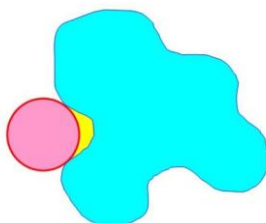


Figure 2 2D-representation of a solvent-excluded surface

1.4 Solvent-accessible surfaces

Solvent-accessible surfaces are obtained in the same way as solvent-excluded surfaces but the outer surface of the solvent sphere is used to define the molecular surface, as shown in Figure 3.

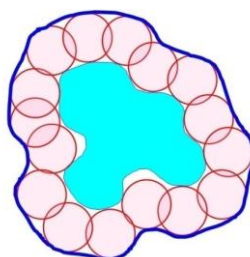


Figure 3 The solvent-accessible surface is obtained by rolling a spherical "solvent molecule"

1.5 Shrink-wrap surface algorithm

Shrink-wrap surface algorithms [3] are used to determine single-valued molecular surfaces. Single-valued in this case means that for any given radial vector from the centre of the molecule the surface is only crossed once (vectors **A** and **B** in Figure 4) and not multiply (vectors **C** and **D** in Figure 4):

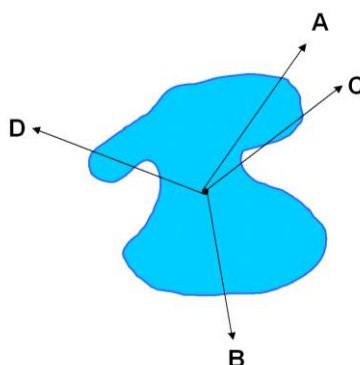


Figure 4 2D-representation of a molecular surface with single-valued (A and B) and multiply valued (C and D) radial vectors from the centre



Single-valued surfaces are necessary for spherical-harmonic fitting (see 1.4). Thus, spherical-harmonic fitting is only available for shrink-wrap surfaces in ParaSurf™. The shrink-wrap algorithm works by starting outside the molecule (point **a** in Figure 5) and moving inwards along the radial vector until it finds the surface (in our case defined by the predefined level of the electron density, point **b** in Figure 5). Thus, the shrink-wrapped surface may contain areas (marked by dashed lines in Figure 5) for which the surface deviates from the true isodensity surface.

These areas of the surface, however, often have little consequence as they are situated above indentations in the molecule that are poorly accessible to solvents or other molecules. The shrink-wrapped surfaces generated by ParaSurf™ should normally be fitted to a spherical-harmonic series for use in HTVS, similarity, pattern-recognition or high-throughput docking applications. The default molecular centre in ParaSurf™ is the centre of gravity (CoG). In special cases in which the CoG lies outside the molecule, another centre may be chosen.

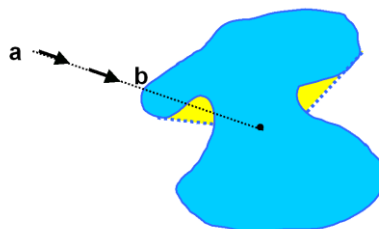


Figure 5 2D-representation of the shrink-wrap algorithm. The algorithm scans along the vector from point **a** towards the centre of the molecule until the electron density reaches the preset value (point **b**). The algorithm results in enclosures (marked yellow) for multi-valued radial vectors

Figure 6 shows a spherical-harmonically fitted shrink-wrap surface for a difficult molecule. The areas shown schematically in Figure 5 are clearly visible.

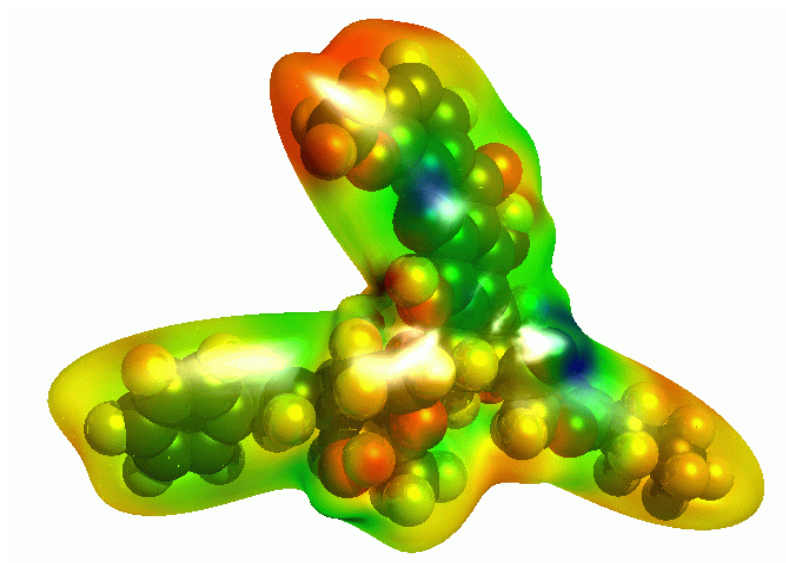


Figure 6 Spherical-harmonic approximation of a shrink-wrap isodensity surface. Note the areas where the surface does not follow the indentations of the molecule



1.6 Marching-cube algorithm

The marching-cube algorithm [4] implemented in ParaSurf™ does not have the disadvantage of being single-valued like the shrink-wrap surface. It cannot, therefore, be fitted to a spherical harmonic series and is used as a purely numerical surface primarily for QSPR applications or surface-integral models. [10] The algorithm works by testing the electron density at the corners of cubes on a cubic lattice laid out through the molecular volume. The corners are divided into those “inside” the molecule (i.e. with a higher electron density than the preset value) and those “outside”. The surface triangulation is then generated for each surface cube and the positions of the surface points corrected to the preset electron density.

1.7 Spherical-harmonic fitting

Complex surfaces can be fitted to spherical harmonic series to give analytical approximations of the surface. [5] The surfaces are fit to a series of distances $r_{\alpha,\beta}$ from the centre along the radial vector defined by the angles α and β as:

$$r_{\alpha,\beta} = \sum_{l=0}^N \sum_{m=-l}^l c_l^m Y_l^m \quad (1)$$

Where the distances $r_{\alpha,\beta}$ are linear combinations of spherical harmonics Y_l^m defined as:

$$Y_l^m(\alpha, \beta) = \sqrt{\frac{(2l+1)(l-m)!}{4\pi(l+m)!}} P_l^m(\cos \alpha) e^{im\beta} \quad (2)$$

where $P_l^m(\cos \alpha)$ are associated Legendre functions and l and m are integers such that $-l \leq m \leq l$. In the above form, spherical harmonics are complex functions. Duncan and Olson [14] have used the real functions

$$Y_l^m(\alpha, \beta) = N_{lm} P_l^m(\cos \alpha) \cos |m| \beta \quad (3)$$

where N_{lm} are normalization factors, to describe molecular surfaces using spherical harmonics.

ParaSurf™ not only fits the surface itself (i.e. the radial distances) to spherical harmonic expansions, but also the four local properties (see 1.8). In this way, a completely analytical description of the shape of the molecule and its intermolecular binding properties is obtained. [15] This description can be truncated at different orders l depending on the application and the precision needed. Thus, a simple description of the molecular properties (shape, MEP, IE_L, EA_L and α_L) to order 2 consists of only five sets of nine coefficients each, or 45 coefficients. These coefficients can be rotated, overlaps calculated etc. [5] to give fast scanning of large numbers of compounds.



Note that, because of the approximate nature of the spherical-harmonic fits, the default isodensity level for the shrink-wrapped surface ($0.00002 \text{ e}^{-\text{\AA}^{-3}}$) is lower than that ($0.0003 \text{ e}^{-\text{\AA}^{-3}}$) appropriate for an approximately van der Waals' surface using the marching-cube algorithm. The lower value avoids the surface coming too close to atoms. Note also that the fits are incremental, which means that the order chosen for a given application can be obtained by ignoring coefficients of higher order in the spherical-harmonic series.

In some cases, the default resolution of the molecular surface does not allow fitting the spherical-harmonic expansion to very high orders without introducing noise ("ripples") on the fitted surface. In this case, the calculated RMSD becomes larger at higher orders of the spherical-harmonic expansion. ParaSurf10™ recognizes this condition and truncates the fitting procedure at the optimum value. This can be recognized in the output because the RMSD for later cycles remains constant and the coefficients of the higher order spherical harmonics are all zero. This guarantees the optimum fit in each case and is important for applications that use either the spherical-harmonic coefficients themselves or the hybridization coefficients.

The choice of centre for fitting to a spherical-harmonic expansion is critical. ParaSurf10™ therefore goes through a multi-step procedure in order to find a suitable centre. This procedure is retained for all molecules for which the ParaSurf08™ found a suitable centre. However, if the algorithms implemented in ParaSurf08™ fail to find a suitable centre, the additional technique implemented in ParaSurf10™ will probably work.

The problem with many molecules is that, for instance, the centre of mass does not lie within the molecular volume. This can easily be the case for, for instance, U- or L-shaped molecules. The procedure implemented in ParaSurf10™ works as follows:

1. The program first calculates the centre of mass and tests whether it lies within the volume of the molecule. If it does, it is used as the molecular centre. If not, the program moves on to the next step.
2. ParaSurf™ calculates the principal moments of inertia of the molecule and derives a centre from them by assuming that the molecule is U- or V-shaped. The procedure tries to place the centre at the base centre of the molecule. This procedure was implemented in ParaSurf08™ as a fallback if the centre of mass proved unsuitable. If it also fails to find a suitable centre, ParaSurf10™ moves on to a third option, which finds a centre for all but the most difficult molecules.
3. The new procedure first searches for the largest plane in the molecule (i.e. the one that contains the most atoms). This search has some leeway, so that the atoms must not all lie exactly in the plane. As a second step, the second largest plane is sought. The molecular centre is then placed in the hinge area between the two planes, as illustrated in Figure 7:

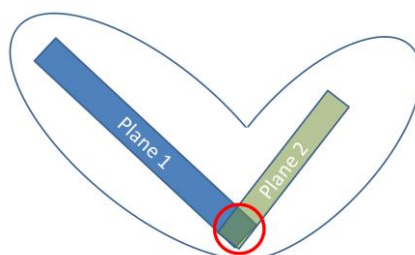


Figure 7 Schematic representation of the planes and hinge area used to determine the centre for spherical-harmonic expansions



1.8 Local properties

The local properties calculated by ParaSurf™ are those related to intermolecular interactions. Local properties, sometimes inaccurately called fields in QSAR work, are properties that vary in space around the molecule and therefore have a distribution of values at the molecular surface. The best known and most important local property in this context is the molecular electrostatic potential, which governs Coulomb interactions, but the MEP only describes a part of the intermolecular interaction energy, so that further local properties are needed.

1.8.1 Molecular electrostatic potential

The MEP is defined in ParaSurf™ as the energy of interaction of a single positive electronic charge at the position $\mathbf{r}$ with the molecule. Within quantum mechanical (semiempirical or *ab initio* molecular orbital (MO) theory, density functional theory (DFT)) the MEP ($V(\mathbf{r})$) is described [6] as:

$$MEP(\mathbf{r}) = \sum_{i=1}^n \frac{Z_i}{|\mathbf{R}_i - \mathbf{r}|} - \int \frac{\rho(\mathbf{r}') d\mathbf{r}'}{|\mathbf{r}' - \mathbf{r}|} \quad (4)$$

where n is the number of atoms in the molecule, Z_i is the nuclear charge of atom i located at $\mathbf{R}_i$ and $\rho(\mathbf{r})$ is the electron-density function of the molecule. This expression, however, involves integrating the electron density, a time-consuming calculation. ParaSurf™ therefore uses two different approximate models for calculating the MEP.

1.8.1.1 The natural atomic orbital/PC (NAO-PC) model

The NAO-PC model [16, 17] uses a total of nine point charges, one positive charge at the nucleus and eight negative ones distributed around it, to describe the electrostatics of a non-hydrogen atom with a valence-only *s*- and *p*-basis set for the semiempirical Hamiltonians MNDO, [18] AM1 [19] and PM3. [20] The negative charges are located at the charge centres of each lobe of the natural atomic orbitals, which are obtained by diagonalising the one-atom blocks of the density matrix. [16] The NAO-PC charges are calculated by VAMP and output in the .sdf file for use in ParaSurf™. The NAO-PC model is therefore only available when using ParaSurf™ with VAMP .sdf input. NAO-PC charges are also not available for semiempirical Hamiltonians such as MNDO/d [21] or AM1* [22] that use *d*-orbitals in the basis set.

1.8.1.2 The multipole model

The integrals needed to evaluate Eq. 4 in MNDO-type methods use a multipole approximation [18, 21] that extends to quadrupoles. We can therefore also use this approximation to calculate atom-centred monopoles, dipoles and quadrupoles for each atom in the molecule. [23] This multipole model is applicable to all methods, including those with *d*-orbitals, and can be used with MOPAC output files as input to ParaSurf™.



1.8.2 Local ionization energy, electron affinity, hardness and electronegativity

The local ionization energy $IE_L(\mathbf{r})$ is defined [7] as a density-weighted Koopmans' ionization potential at a point $\mathbf{r}$ near the molecule:

$$IE_L(\mathbf{r}) = \frac{-\sum_{i=1}^{HOMO} \rho_i(\mathbf{r}) \varepsilon_i}{\sum_{i=1}^{HOMO} \rho_i(\mathbf{r})} \quad (5)$$

where $HOMO$ is the number of the highest occupied MO, $\rho_i(\mathbf{r})$ is the electron density at point $\mathbf{r}$ due to MO i and ε_i is its Eigenvalue. The local ionization energy describes the tendency of the molecule to interact with electron acceptors (Lewis acids) in a given region in space. [7, 8]

The definition of the local electron affinity is a simple extension of Eq. 5 to the virtual MOs: [8]

$$EA_L(\mathbf{r}) = \frac{-\sum_{i=LUMO}^{norbs} \rho_i(\mathbf{r}) \varepsilon_i}{\sum_{i=LUMO}^{norbs} \rho_i(\mathbf{r})} \quad (6)$$

The local electron affinity is the equivalent of the local ionization energy for interactions with electron donors (Lewis bases). [8]

Eq. 6 requires that the occupied and virtual orbitals be approximately equivalent to each other. This is not the case for semiempirical Hamiltonians (such as AM1*) that include d -orbitals as polarisation functions or for extensive basis sets in Hartree-Fock *ab initio* or in Density-Functional theory (DFT) calculations. A new technique has therefore been defined [11] to exclude pure polarisation functions from the summation in Eq. 6. This technique is now default in ParaSurf 10™ and gives reliable results. For continuity, a new command-line option (EAL09) has been introduced to request that the calculation of the local electron affinity be performed exactly as in ParaSurf 09™ and earlier versions.

Two further, less fundamental local properties have been defined. [8] These are the local hardness, η_L :

$$\eta_L = \frac{(IE_L - EA_L)}{2} \quad (7)$$

and the local electronegativity, χ_L :

$$\chi_L = \frac{(IE_L + EA_L)}{2} \quad (8)$$



1.8.3 Local polarisability

Within the NDDO, the molecular electronic polarisability is easily accessible using the parameterized version [24] of the variational technique introduced by Rivail, [25] which can also be partitioned into an additive polarisability scheme. [26] This allows us to define the local polarisability, α_L , at a point near the molecule as

$$\alpha_L(\mathbf{r}) = \frac{\sum_{j=1}^{norbs} \rho_j^1(\mathbf{r}) q_j \bar{\alpha}_j}{\sum_{j=1}^{norbs} \rho_j^1(\mathbf{r}) q_j} \quad (9)$$

where q_j is the Coulson occupation and $\bar{\alpha}_j$ the isotropic polarisability attributed to atomic orbital j . The density ρ_j^1 is defined as the electron density at the point in question due to an exactly singly occupied atomic orbital j . The sum is now over atomic orbitals, rather than MOs as for the other local properties. Thus, the local polarisability is a simple occupation-weighted sum of the orbital polarisabilities in which the contribution of each AO is determined by the density of the individual AO at the point being considered.

1.8.4 Field normal to the surface

The electrostatic field (the first derivative of the potential) normal to the molecular surface is closely related to the electrostatic solvation energy in implicit solvation models. [27, 28] This field also has the advantage that it is largely independent of the total molecular charge, so that charged molecules can be compared with neutral ones. If the molecular electrostatic potential is used for this purpose, the charge of ions leads a shift in the potential descriptors, so that molecules and ions with different charges cannot be compared directly. The direction of the normal field (inwards or outwards) also defines, for instance hydrogen-bond donors and acceptors specifically.



1.9 Descriptors

A set of 81 molecular descriptors derived from the MEP, local ionization energy, IE_L , electron affinity, EA_L , electronegativity, χ_L , hardness, η_L , and polarisability, α_L has been defined for QSPR-studies. [9] These and several related descriptors calculated and output by ParaSurf™ are defined in the following table.

Table 1: The descriptors calculated by ParaSurf™

Descriptor	Description	Formula/ Reference	Symbol in CSV file
μ	Dipole moment		dipole
μ_D	Dipolar density	[21]	dipden
α	Molecular electronic polarisability	[29]	polarizability
MW	Molecular weight		MWt
G	Globularity	[30]	globularity
A	Molecular surface area		totalarea
VOL	Molecular volume		volume
$V_{\max}$	Maximum (most positive) MEP	[31]	MEPmax
$V_{\min}$	Minimum (most negative) MEP	[31]	MEPmin
$\bar{V}_+$	Mean of the positive MEP values	[31]	meanMEP+
$\bar{V}_-$	Mean of the negative MEP values	[31]	meanMEP-
$\bar{V}$	Mean of all MEP values	[31]	meanMEP
ΔV	MEP-range	[31]	MEP-range
σ_+^2	Total variance in the positive MEP values	[31]	MEPvar+
σ_-^2	Total variance in the negative MEP values	[31]	MEPvar-
σ_{tot}^2	Total variance in the MEP	[31]	MEPvartot
v	MEP balance parameter	[31]	MEPbalance
$\sigma_{tot}^2 v$	Product of the total variance in the MEP and the balance parameter	[31]	var*balance
γ_1^V	Skewness of the MEP-distribution	$\gamma_1^{\alpha_L} = \frac{\sum_{i=1}^N (\alpha_L^i - \bar{\alpha}_L)^3}{(N-1)\sigma^3}$	MEPskew



Descriptor	Description	Formula/ Reference	Symbol in CSV file
γ_2^V	Kurtosis of the MEP-distribution	$\gamma_2^V = \frac{\sum_{i=1}^N (V_i - \bar{V})^4}{(N-1)\sigma^4} - 3$	MEPkurt
$\int_V$	Integrated MEP over the surface	$\int_V = \sum_{i=1}^N V_i a_i$	MEPint
$IE_L^{\max}$	Maximum value of the local ionization energy		IELmax
$IE_L^{\min}$	Minimum value of the local ionization energy		IELmin
$\overline{IE_L}$	Mean value of the local ionization energy	$\overline{IE_L} = \frac{1}{N} \sum_{i=1}^N IE_L^i$	IELbar
ΔIE_L	Range of the local ionization energy	$\Delta IE_L = IE_L^{\max} - IE_L^{\min}$	IELrange
σ_{IE}^2	Variance in the local ionization energy	$\sigma_{IE}^2 = \frac{1}{N} \sum_{i=1}^N [IE_L^i - \overline{IE_L}]^2$	IELvar
$\gamma_1^{IE_L}$	Skewness of the local ionization energy distribution	$\gamma_1^{IE_L} = \frac{\sum_{i=1}^N (IE_L^i - \overline{IE_L})^3}{(N-1)\sigma^3}$	IELskew
$\gamma_2^{IE_L}$	Kurtosis of the local ionization energy distribution	$\gamma_2^{IE_L} = \frac{\sum_{i=1}^N (IE_L^i - \overline{IE_L})^4}{(N-1)\sigma^4} - 3$	IELkurt
$\int_{IE_L}$	Integrated local ionization energy over the surface	$\int_{IE_L} = \sum_{i=1}^N IE_L^i a_i$	IELint
$EA_L^{\max}$	Maximum of the local electron affinity		EALmax
$EA_L^{\min}$	Minimum of the local electron affinity		EALmin
$\overline{EA_{L+}}$	Mean of the positive values of the local electron affinity	$\overline{EA_{L+}} = \frac{1}{N^+} \sum_{i=1}^{N^+} EA_{L+}^i$	EALbar+
$\overline{EA_{L-}}$	Mean of the negative values of the local electron affinity	$\overline{EA_{L-}} = \frac{1}{N^-} \sum_{i=1}^{N^-} EA_{L-}^i$	EALbar-
$\overline{EA_L}$	Mean value of the local electron affinity	$\overline{EA_L} = \frac{1}{N} \sum_{i=1}^N EA_L^i$	EALbar
ΔEA_L	Range of the local electron affinity	$\Delta EA_L = EA_L^{\max} - EA_L^{\min}$	EALrange



Descriptor	Description	Formula/ Reference	Symbol in CSV file
σ_{EA+}^2	Variance in the local electron affinity for all positive values	$\sigma_{EA+}^2 = \frac{1}{m} \sum_{i=1}^m \left[EA_i^+ - \overline{EA^+} \right]^2$	EALvar+
σ_{EA-}^2	Variance in the local electron affinity for all negative values	$\sigma_{EA-}^2 = \frac{1}{n} \sum_{i=1}^n \left[EA_i^- - \overline{EA^-} \right]^2$	EALvar-
σ_{EAtot}^2	Sum of the positive and negative variances in the local electron affinity	$\sigma_{EAtot}^2 = \sigma_{EA+}^2 + \sigma_{EA-}^2$	EALvartot
ν_{EA}	Local electron affinity balance parameter	$\nu_{EA} = \frac{\sigma_{EA+}^2 \cdot \sigma_{EA-}^2}{\left[\sigma_{EA}^2 \right]^2}$	EALbalance
δA_{EA}^+	Fraction of the surface area with positive local electron affinity	$\delta A_{EA}^+ = \frac{A_{EA}^+}{A}$, A = total surface area	EALfraction+
A_{EA}^+	Surface area with positive local electron affinity		EALarea+
$\gamma_1^{EA_L}$	Skewness of the local electron affinity distribution	$\gamma_1^{EA_L} = \frac{\sum_{i=1}^N (EA_L^i - \overline{EA_L})^3}{(N-1)\sigma^3}$	EALskew
$\gamma_2^{EA_L}$	Kurtosis of the local electron affinity distribution	$\gamma_2^{EA_L} = \frac{\sum_{i=1}^N (EA_L^i - \overline{EA_L})^4}{(N-1)\sigma^4} - 3$	EALkurt
$\int_{EA_L}$	Integrated local electron affinity over the surface	$\int_{EA_L} = \sum_{i=1}^N EA_L^i a_i$	EALint
$\alpha_L^{\max}$	Maximum value of the local polarisability		POLmax
$\alpha_L^{\min}$	Minimum value of the local polarisability		POLmin
$\overline{\alpha_L}$	Mean value of the local polarisability	$\overline{\alpha_L} = \frac{1}{N} \sum_{i=1}^N \alpha_L^i$	POLbar
$\Delta \alpha_L$	Range of the local polarisability	$\Delta \alpha_L = \alpha_L^{\max} - \alpha_L^{\min}$	POLrange
σ_{α}^2	Variance in the local polarisability	$\sigma_{\alpha}^2 = \frac{1}{N} \sum_{i=1}^N \left[\alpha_L^i - \overline{\alpha_L} \right]^2$	POLvar
$\gamma_1^{\alpha_L}$	Skewness of the local polarisability distribution	$\gamma_1^{\alpha_L} = \frac{\sum_{i=1}^N (\alpha_L^i - \overline{\alpha_L})^3}{(N-1)\sigma^3}$	POLskew



Descriptor	Description	Formula/ Reference	Symbol in CSV file
$\gamma_2^{\alpha_L}$	Kurtosis of the local polarisability distribution	$\gamma_2^{\alpha_L} = \frac{\sum_{i=1}^N (\alpha_L^i - \bar{\alpha}_L)^4}{(N-1)\sigma^4} - 3$	POLkurt
$\int_{\alpha_L}$	Integrated local polarisability over the surface	$\int_{\alpha_L} = \sum_{i=1}^N \alpha_L^i a_i$	POLint
$\chi_L^{\max}$	Maximum value of the local electronegativity		ENEGmax
$\chi_L^{\min}$	Minimum value of the local electronegativity		ENEGmin
$\overline{\chi_L}$	Mean value of the local electronegativity	$\overline{\chi_L} = \frac{1}{N} \sum_{i=1}^N \chi_L^i$	ENEGbar
$\Delta\chi_L$	Range of the local electron electronegativity	$\Delta\chi_L = \chi_L^{\max} - \chi_L^{\min}$	ENEGrange
σ_{χ}^2	Variance in the local electronegativity	$\sigma_{\chi}^2 = \frac{1}{N} \sum_{i=1}^N \left[\chi_L^i - \overline{\chi_L} \right]^2$	ENEGvar
$\gamma_1^{\chi_L}$	Skewness of the local electronegativity distribution	$\gamma_1^{\chi_L} = \frac{\sum_{i=1}^N (\chi_L^i - \bar{\chi}_L)^3}{(N-1)\sigma^3}$	ENEGskew
$\gamma_2^{\chi_L}$	Kurtosis of the local electronegativity distribution	$\gamma_2^{\chi_L} = \frac{\sum_{i=1}^N (\chi_L^i - \bar{\chi}_L)^4}{(N-1)\sigma^4} - 3$	ENEGkurt
$\int_{\chi_L}$	Integrated local electronegativity over the surface	$\int_{\chi_L} = \sum_{i=1}^N \chi_L^i a_i$	ENEGint
$\eta_L^{\max}$	Maximum value of the local hardness		HARDmax
$\eta_L^{\min}$	Minimum value of the local hardness		HARDmin
$\overline{\eta_L}$	Mean value of the local hardness	$\overline{\eta_L} = \frac{1}{N} \sum_{i=1}^N \eta_L^i$	HARDbar
$\Delta\eta_L$	Range of the local electron hardness	$\Delta\eta_L = \eta_L^{\max} - \eta_L^{\min}$	HARDrange
σ_{η}^2	Variance in the local hardness	$\sigma_{\eta}^2 = \frac{1}{N} \sum_{i=1}^N \left[\eta_L^i - \overline{\eta_L} \right]^2$	HARDvar
$\gamma_1^{\eta_L}$	Skewness of the local hardness distribution	$\gamma_1^{\eta_L} = \frac{\sum_{i=1}^N (\eta_L^i - \bar{\eta}_L)^3}{(N-1)\sigma^3}$	HARDskew



Descriptor	Description	Formula/ Reference	Symbol in CSV file
$\gamma_2^{\eta_L}$	Kurtosis of the local hardness distribution	$\gamma_2^{\eta_L} = \frac{\sum_{i=1}^N (\eta_L^i - \bar{\eta}_L)^4}{(N-1)\sigma^4} - 3$	HARDkurt
$\int_{\eta_L}$	Integrated local hardness over the surface	$\int_{\eta_L} = \sum_{i=1}^N \eta_L^i a_i$	HARDint
$F_N^{\max}$	Maximum value of the electrostatic field normal to the surface		FNmax
$F_N^{\min}$	Minimum value of the field normal to the surface		FNmin
$\overline{F_N}$	Mean value of the field normal to the surface	$\overline{F_N} = \frac{1}{N} \sum_{i=1}^N \chi_L^i$	FNmean
σ_F^2	Variance in field normal to the surface	$\sigma_F^2 = \frac{1}{N} \sum_{i=1}^N \left[F_N^i - \overline{F_N} \right]^2$	FNvartot
σ_{F+}^2	Variance in the field normal to the surface for all positive values	$\sigma_{F+}^2 = \frac{1}{m} \sum_{i=1}^m \left[F_N^{i+} - \overline{F_N^+} \right]^2$	FNvar+
σ_{F-}^2	Variance in the field normal to the surface for all negative values	$\sigma_{F-}^2 = \frac{1}{n} \sum_{i=1}^n \left[F_N^{i-} - \overline{F_N^-} \right]^2$	FNvar-
ν_F	Normal field balance parameter	$\nu_F = \frac{\sigma_{F+}^2 + \sigma_{F-}^2}{\left[\sigma_F^2 \right]^2}$	FNbal
$\gamma_1^{F_N}$	Skewness of the field normal to the surface	$\gamma_1^{F_N} = \frac{\sum_{i=1}^N (F_N^i - \overline{F_N})^3}{(N-1)\sigma^3}$	FNskew
$\gamma_2^{F_N}$	Kurtosis of the field normal to the surface	$\gamma_2^{F_N} = \frac{\sum_{i=1}^N (F_N^i - \overline{F_N})^4}{(N-1)\sigma^4} - 3$	FNkurt
$\int_{F_N}$	Integrated field normal to the surface over the surface	$\int_{F_N} = \sum_{i=1}^N F_N^i a_i$	FNint
$\int_{F_N}^+$	Integrated field normal to the surface over the surface for all positive values	$\int_{F_N}^+ = \sum_{i=1}^N F_N^i a_i \text{ if } F_N^i \geq 0$	FN+
$\int_{F_N}^-$	Integrated field normal to the surface over the surface for all negative values	$\int_{F_N}^- = \sum_{i=1}^N F_N^i a_i \text{ if } F_N^i < 0$	FN-



Descriptor	Description	Formula/ Reference	Symbol in CSV file
$\int_{ F_N }$	Integrated absolute field normal to the surface over the surface	$\int_{F_N} = \sum_{i=1}^N F_N^i a_i$	FNabs
Additionally if the Shannon Entropy is calculated			
$H_{in}^{\max}$	Maximum value of the internal Shannon Entropy		SHANImax
$H_{in}^{\min}$	Minimum value of the internal Shannon Entropy		SHANImin
$\overline{H_{in}}$	Mean value of the internal Shannon Entropy	$\overline{H_{in}} = \frac{1}{N} \sum_{i=1}^N H_{in}^i$	SHANIbar
$\sigma_{H_{in}}^2$	Variance in the internal Shannon Entropy	$\sigma_{H_{in}}^2 = \frac{1}{N} \sum_{i=1}^N \left[H_{in}^i - \overline{H_{in}} \right]^2$	SHANIvar
$\int_{H_{in}}$	Integrated internal Shannon Entropy over the surface	$\int_{H_{in}} = \sum_{i=1}^N H_{in}^i a_i$	SHANItot
And if the external Shannon Entropy is available			
$H_{ex}^{\max}$	Maximum value of the external Shannon Entropy		SHANEmax
$H_{ex}^{\min}$	Minimum value of the external Shannon Entropy		SHANEmin
$\overline{H_{ex}}$	Mean value of the external Shannon Entropy	$\overline{H_{ex}} = \frac{1}{N} \sum_{i=1}^N H_{ex}^i$	SHANEbar
$\sigma_{H_{ex}}^2$	Variance in the external Shannon Entropy	$\sigma_{H_{ex}}^2 = \frac{1}{N} \sum_{i=1}^N \left[H_{ex}^i - \overline{H_{ex}} \right]^2$	SHANEvar
$\int_{H_{ex}}$	Integrated internal Shannon Entropy over the surface	$\int_{H_{ex}} = \sum_{i=1}^N H_{ex}^i a_i$	SHANEtot

1.10 Surface-integral models (polynomial version)*

1.11 Binned surface-integral models*

* Only available in the full version.



1.12 Spherical harmonic “hybrids”

Once the molecular shape or a local property have been fitted to a spherical-harmonic expansion, [13] the shape or property can be described succinctly as a series of spherical-harmonic “hybridization” coefficients analogous to the concept of hybrid atomic orbitals. Thus, for each value of l in Eq. 1 the “hybridization” coefficient H_l is given by:

$$H_l = \sum_{i=-m}^m (c_l^m)^2 \quad (11)$$

The hybridization coefficients H_l can be used as additional descriptors for fast QSPR screening.

1.13 Descriptors and moments based on polynomial surface-integral models*

1.14 Shannon entropy*

1.15 Surface autocorrelations*

1.16 Standard Rotationally Invariant Fingerprints (RIFs)

Mavridis et al. [32] introduced standard rotationally invariant fingerprints (RIFs) based on the spherical-harmonic hybridization coefficients defined above. These fingerprints provide a detailed description of the molecular shape, electrostatics, donor/acceptor properties and polarisability as a standard series of 54 floating point numbers.

1.17 Maxima and Minima of the Local Properties*

1.18 Atom-centred descriptors*

1.19 Fragment analysis*

* Only available in the full version.



2 PROGRAM OPTIONS

2.1 Command-line options

ParaSurf™ program options are given as command-line arguments. Arguments are separated by blanks, so that no single argument may contain a blank character. Arguments may be written in any combination of upper and lower case. The options are:

Table 2: ParaSurf™ command-line options

<name>		Base name for the input file (must be the first argument). <name> is not required if the first argument is -version (see below)
		Using this option, the input file is assumed to be <name>_v.sdf if a file with this name exists.
		Otherwise the file <name>.sdf will be used as input.
		If neither of these files are found, the program will use an .sdf file written by the Cepos version of Mopac 6. These files are called <name>_m.sdf
		The output files are <name>_p.out <name>_p.sdf <name>.psf (optional)* <name>.asd (optional)* <name>_p.vmp (optional)*
surf=	wrap cube	Shrink-wrap surface (default) Marching-cube surface
contour=	isoden solvex	The surface is defined by the electron density A solvent-excluded surface is used.
fit=	sphh isod none	Spherical-harmonic fitting (default for surf=wrap) Smooth to preset isodensity value (default for surf=cube) No fitting
iso=	n.nn	Isodensity value set to n.nn e ⁻³ (default for shrink-wrap surface = 0.00002 ; default for marching-cube surface = 0.0003 ; minimum possible value = 0.00001)

* Only available in the full version.



rsol=	<i>n.nn</i>	A solvent-probe radius of <i>n.nn</i> Å is used for calculating the solvent-excluded or solvent-accessible surface (default= 1.0 , allowed range is from 0.0 to 2.0 Å)
mesh=	<i>n.nn</i>	The mesh size used to triangulate the surface is set to <i>n.nn</i> Å (default value = 0.2 Å, allowed range is from 0.1 to 1.0 Å)
grid=	<filename> auto	Read the Cartesian coordinates at which to calculate a grid of the four properties (MEP, IE _L , EA _L , α _L). See 3.8.1 ParaSurf™ calculates an automatic grid (see 3.8.2)
lattice=	<i>n.nn</i>	Sets the lattice spacing for the grid=auto option (see 3.8.2)
-version		Must be the first argument. Requests that ParaSurf™ prints the version number to the standard output channel and then stops without performing a calculation.
eal09		Do not use the selection procedure for virtual orbitals [11] when calculating the local electron affinity. This option provides continuity with earlier versions of ParaSurf™

Examples:

```
parasurf test surf=wrap fit=sphh iso=0.03 estat=naopc
```

Use the input file **test_v.sdf**, **test.sdf** or **test_m.sdf** to calculate a shrink-wrap surface with an isodensity value of $0.03 \text{ e}^{-\text{\AA}^{-3}}$, perform a spherical-harmonic fit, use NAO-PC electrostatics and write the spherical-harmonic coefficients to **test_P.sdf**.

```
parasurf test surf=cube fit=none
```

Use the file **test_v.sdf**, **test.sdf** or **test_m.sdf** as input to perform a marching-cube surface determination without fitting and to calculate the descriptor set.



2.2 Options defined in the input SDF-file

2.2.1 Defining the centre for spherical-harmonic fits

The automatic determination of the molecular centre for spherical-harmonic fitting can be overridden by adding a field to the Input (usually VAMP) SDF-file with the tag:

<SPHH_CENTER>

The centre can be defined using Cartesian coordinates using an input line (immediately after the **SPHH_CENTER** tag) of the format:

Cartesian x.xx y.yy z.zz

where **x.xx**, **y.yy** and **z.zz** are the x, y, and z-coordinates, respectively. The capitalization of "Cartesian" is required.

Alternatively, a list of atoms can be given using the format

Atoms n1 n2 n3 n4 n5 n6 ...

where **n1** etc. are the numbers of the atoms to be used to calculate the centre of gravity. The capitalization of "Atoms" is required and the list of atoms is limited to one line.

2.2.2 Defining fragments^{*}

^{*} Only available in the full version.



3 INPUT AND OUTPUT FILES

ParaSurf™ uses the following files for input and output:

Table 3: ParaSurf™ input and output files

File	Name	Description
Input	<filename>_v.sdf or <filename>.sdf (if available) or <filename>_m.sdf	VAMP .sdf file output. VAMP must be run with the ALLVECT option to be able to calculate all the properties. The VAMP version used must be able to calculate AO-polarisabilities. If no VAMP .sdf file is found, ParaSurf™ defaults to a Cepas Mopac 6 .sdf file. It is strongly recommended to use the EF option for geometry optimizations in Mopac.
Hamiltonian	Vhamil.par	The VAMP parameters file (also found in the VAMP executable directory). This file must be copied to the ParaSurf™ executable directory.
Output	<filename>_p.out	Always written.
SD-file	<filename>_p.sdf	Always written.

3.1 The VAMP .sdf file as input

VAMP .sdf files, an extension of the MDL .sdf file format, [33] are the primary communication channel between VAMP and ParaSurf™. The atomic coordinates and bond definitions are given in the MDL format as shown in **Figure 8**. The remaining fields are indicated by tags with the form:

<FIELD_NAME>

FIELD_NAME is a predefined text tag used to locate the relevant data within the .sdf file.

Only the important fields for a ParaSurf™ calculation will be described here:

<HAMILTONIAN>

The Hamiltonian field defines the semiempirical Hamiltonian (model and parameters) used for the calculation. The Hamiltonian must be defined for ParaSurf™ to be able to calculate the electrostatics and the local polarisabilities. NAO-PC electrostatics and the local polarisability are not available for all methods. Quite generally, the multipole electrostatics model is to be preferred over the NAO-PC model, which can only be used if the VAMP .sdf file contains a block with the tag:

<NAO-PC>

NAO-PCs cannot be calculated for methods with *d*-orbitals. The local polarisability calculation has not yet been extended to these methods, but will be in a future release.



```

1-Bromo-3,5-difluorobenzene
OMVAMP81A04250313563D 1 0.00000 0.00000 0

12 12 0 0 0 0 1 V2000
-2.6274 0.2410 0.0003 F
-1.2738 0.2410 0.0003 C
-0.5810 1.4623 0.0003 C
0.8231 1.4389 0.0003 C
1.5096 2.6055 0.0004 F
1.5266 0.2198 0.0001 C
0.8142 -0.9793 0.0001 C
1.7431 -2.6055 -0.0004 Br
-0.5805 -0.9840 0.0002 C
-1.1264 2.4167 -0.0003 H
2.6274 0.2339 0.0003 H
-1.1515 -1.9253 0.0001 H
1 2 1
2 3 4
3 4 4
4 5 1
4 6 4
6 7 4
7 8 1
2 9 4
7 9 4
3 10 1
6 11 1
9 12 1
M END

```

Figure 8 The headers and titles, atomic coordinates and bond definitions from a VAMP .sdf file. The format follows the MDL definition [25]

The following table gives an overview of the methods and their limitations:

Table 4: Hamiltonians and the available electrostatic and polarisability models.

Hamiltonian	Reference	Electrostatics		Local Polarisability
		NAO-PC	Multipole	
MNDO	[18]	YES	YES	YES
AM1	[19]	YES	YES	YES
PM3	[20]	YES	YES	YES
MNDO/c	[34]	YES	YES	NO
MNDO/d	[21]	NO	YES	NO
AM1*	[22]	NO	YES	NO

<VAMPBASICS>

The VAMPBASICS block contains the following quantities (FORTRAN format 6f13.6):

Heat of Formation	kcal mol ⁻¹
HOMO energy	eV
LUMO energy	eV
Dipole moment	



x-component	Debye
y-component	Debye
z-component	Debye

<TOTAL COULSON CHARGE>

The total charge of the molecule.

<DENSITY MATRIX ELEMENTS>

The DENSITY MATRIX ELEMENTS block contains the one-atom blocks of the density matrix for the non-hydrogen atoms. For an *sp*-atom, there are ten elements, for an *spd*-atom 45. The squares of the diagonal elements for hydrogen atoms are included in the <CHARGE ON HYDROGENS> block that follows the density matrix. The density-matrix elements are used in ParaSurf™ to calculate the local properties and are essential.

<ORBITAL VECTORS>

The ORBITAL VECTORS block contains the MO-eigenvectors and related information and is essential for calculating the local properties. VAMP must be run with the keyword **ALLVECT** in order to write all the MO vectors to the SDF file.

The entire SDF input file is echoed to the <filename>_p.sdf output file and the properties calculated by ParaSurf™ are added in additional blocks at the end.

3.1.1 Multi-structure SD-files*

3.2 The Cepos MOPAC 6.sdf file as input

Cepos Mopac 6 writes an .sdf file containing the above blocks with the exception that the MOPACBASICS block replaces VAMPBASICS. No additional keywords are required to request the correct .sdf output for ParaSurf™.

3.3 The Vhamil.par file

The file Vhamil.par is used by VAMP to define the available Hamiltonians and elements and supply the parameters. This file is also used by ParaSurf™ for the same purpose. A Vhamil.par file for standard Hamiltonians and elements is supplied with the ParaSurf™ program. In order to be sure that all Hamiltonians and elements available to VAMP can also be handled by ParaSurf™, however, the

* Only available in the full version.



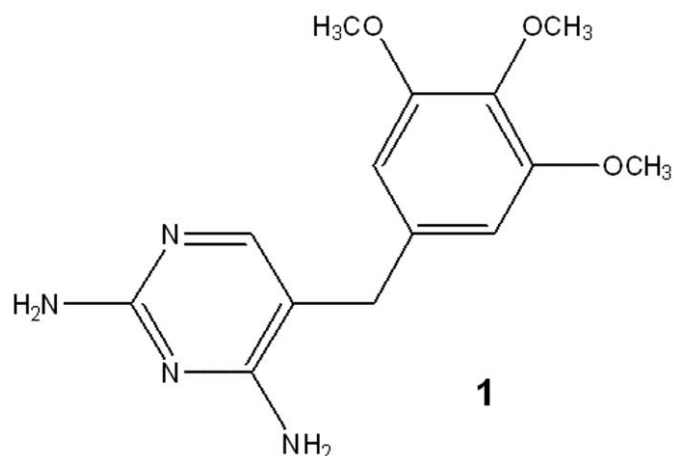
Vhamil.par file from the VAMP executable directory should be copied into the ParaSurf™ executable directory.

3.4 The ParaSurf™ output file

The ParaSurf™ output file provides the user with information about the calculation and the results. It is, however, not intended as the primary means of communication between ParaSurf™ and other programs. Thus, the essential information contained in the output file is also available from the ParaSurf™ output .sdf file.

3.4.1 For a spherical-harmonic surface

Figure 9 shows the output for a calculation using the options **surf=wrap fit=sphh translate** for trimethoprim, **1**.



```
<> ParaSurf'10 Academic Version
<> Copyright (c) 2006,2007,2008,2009,2010 Friedrich-Alexander-Universitaet
    Erlangen-Nuernberg and Cepos InSilico Ltd.
    All rights reserved.

<> Input = trimethoprim.sdf

<> Program options :

    Using shrink-wrap isocontour surface
    Fitting surface to spherical harmonics
    Using an isodensity surface contour
    Isodensity value = 0.2000E-04 electrons/Angstrom**3
    Triangulation mesh = 0.20 Angstrom
    Using multipole electrostatics
```

Figure 9 ParaSurf™ output for trimethoprim, **1**, using a spherical-harmonic surface



Figure 9 continued

```
<> AM1 calculation for Trimethoprim
<> Fitting surface to spherical harmonics
<> Order(l)    RMSD

    0      1.92523465
    1      1.96112137
    2      1.55522693
    3      1.10610407
    4      0.93107725
    5      0.70604021
    6      0.63660643
    7      0.57075868
    8      0.52397917
    9      0.50047987
   10      0.47258810
   11      0.44413396
   12      0.41917572
   13      0.40317390
   14      0.39305683
   15      0.38338670

<> Spherical harmonic fit for MEP:
<> Order(l)    RMSD

    0      11.06679887
    1      11.02896096
    2      8.63328473
    3      6.86212353
    4      5.49789123
    5      4.58515348
    6      4.17133704
    7      3.45021190
    8      3.12551976
    9      2.77785314
   10      2.36026963
   11      2.07229809
   12      1.90436905
   13      1.72364959
   14      1.64559721
   15      1.46827862
   16      1.27844169
   17      1.07448912
   18      0.93472662
   19      0.88271611
   20      0.82766903

<> Spherical harmonic fit for IE(l):
<> Order(l)    RMSD

    0      56.96267883
    1      50.08995849
    2      45.41506823
    3      43.50398342
    4      40.07275301
    5      35.50489297
    6      32.77872159
    7      26.87765240
    8      23.10395662
    9      19.60420788
   10      17.97895259
   11      16.20677848
   12      15.12291903
   13      14.61861765
   14      13.86580806
   15      13.43494797
   16      13.22461242
   17      12.62315996
   18      12.26642783
   19      12.26642783
   20      12.26642783
```



Figure 9 continued

```
<> Spherical harmonic fit for EA(l):
<> Order(l)    RMSD

    0    12.18546245
    1    11.86493015
    2    11.74652208
    3     9.50375558
    4     8.72724156
    5     7.28928968
    6     7.13942468
    7     6.79112856
    8     6.47917898
    9     6.02700721
   10     5.73179346
   11     5.46853066
   12     5.18673696
   13     4.51767269
   14     4.11406977
   15     3.92082708
   16     3.68234845
   17     3.60371852
   18     3.40232841
   19     3.23619071
   20     3.10610529

<> Spherical harmonic fit for Field(N):
<> Order(l)    RMSD

    0     7.95868149
    1     7.94096053
    2     7.17401521
    3     6.65225208
    4     5.99511525
    5     5.55916448
    6     5.41073978
    7     5.06133734
    8     4.86238016
    9     4.56150784
   10     4.11032910
   11     3.75302455
   12     3.52058064
   13     3.43143087
   14     3.35170488
   15     3.08389445
   16     2.71006812
   17     2.33242842
   18     2.10812470
   19     2.03521010
   20     1.94087735

<> Spherical harmonic fit for Alpha(l):
<> Order(l)    RMSD

    0     0.02367600
    1     0.01660891
    2     0.01369187
    3     0.01110787
    4     0.00911432
    5     0.00818939
    6     0.00769335
    7     0.00722758
    8     0.00695207
    9     0.00643405
   10     0.00588639
```



Figure 9 continued

```
11      0.00574100
12      0.00532006
13      0.00531482
14      0.00521057
15      0.00514790
16      0.00514790
17      0.00514790
18      0.00514790
19      0.00514790
20      0.00514790

<> Property ranges:
Density   : 0.3567E-05 to 0.9969E-04
IE(l)     : 391.03 to 671.23
EA(l)     : -108.54 to -38.27
MEP       : -48.51 to 16.80
Alpha(l)  : 0.2369 to 0.3374
Field(N)  : -41.48 to 16.61

<> Descriptors :

Dipole moment      : 1.2486 Debye
Dipolar density    : 0.1936E-02 Debye.Angstrom**-3
Molecular pol.     : 31.2348 Angstrom**3
Molecular weight   : 290.32
Globularity        : 0.7689
Total surface area : 469.52 Angstrom**2
Molecular volume   : 644.95 Angstrom**3

Most positive MEP   : 16.80 kcal/mol
Most negative MEP   : -48.51 kcal/mol
Mean +ve MEP        : 5.59 kcal/mol
Mean -ve MEP        : -10.79 kcal/mol
Mean MEP            : -3.13 kcal/mol
MEP range           : 65.31 kcal/mol
MEP +ve Variance    : 10.80 (kcal/mol)**2
MEP -ve Variance    : 94.42 (kcal/mol)**2
MEP total variance  : 105.22 (kcal/mol)**2
MEP balance parameter: 0.0921
MEP balance*variance : 9.6913 kcal/mol
MEP skewness        : -1.1819
MEP kurtosis        : 1.3877
Integral MEP         : -1166.29 kcal.Angstrom**2/mol

Maximum IE(l)       : 671.23 kcal/mol
Minimum IE(l)       : 391.03 kcal/mol
Mean IE(l)          : 475.67 kcal/mol
IE(l) range         : 280.20 kcal/mol
IE(l) variance      : 3233.34 (kcal/mol)**2
IE(l) skewness      : 0.6768
IE(l) kurtosis      : -0.2286
Integral IE(l)      : 9649.93 eV.Angstrom**2

Maximum EA(l)       : -38.27 kcal/mol
Minimum EA(l)       : -108.54 kcal/mol
Mean +ve EA(l)      : 0.00 kcal/mol
Mean -ve EA(l)      : -93.87 kcal/mol
Mean EA(l)          : -93.87 kcal/mol
EA(l) range         : 70.28 kcal/mol
EA(l) +ve variance  : 0.00 (kcal/mol)**2
EA(l) -ve variance  : 142.43 (kcal/mol)**2
EA(l) total variance : 142.43 (kcal/mol)**2
EA(l) skewness      : 1.7819
EA(l) kurtosis      : 4.1713
Integral EA(l)      : -1913.50 eV.Angstrom**2
EA(l) balance param. : 0.0000
Fraction pos. EA(l) : 1.0000 ( = 469.52 Angstrom**2)
```



Figure 9 continued

```
Max. local Eneg.      :      299.61 kcal/mol
Min. local Eneg.      :      143.17 kcal/mol
Mean local Eneg.      :      190.90 kcal/mol
Local Eneg. range     :      156.44 kcal/mol
Local Eneg. variance  :      958.85 (kcal/mol)**2
Local Eneg. skewness  :          0.82
Local Eneg. kurtosis   :          0.01
Integral local Eneg.   : 3868.21      eV.Angstrom**2

Max. local hardness   :      371.63 kcal/mol
Min. local hardness   :      247.42 kcal/mol
Mean local hardness   :      284.77 kcal/mol
Local hard. range     :      124.21 kcal/mol
Local hard. variance  :      729.04 (kcal/mol)**2
Local hard. skewness  :          0.58
Local hard. kurtosis   :         -0.48
Integral local Hard.   : 5781.71      eV.Angstrom**2

Maximum alpha(l)      :      0.3374      Angstrom**3
Minimum alpha(l)      :      0.2369      Angstrom**3
Mean alpha(l)         :      0.2821      Angstrom**3
Alpha(l) range        :      0.1005      Angstrom**3
Variance in alpha(l)  :      0.5455E-03 Angstrom**6
Alpha(l) skewness     :         -0.7808
Alpha(l) kurtosis     :         -0.6145
Integral Alpha(l)      : 132.761      Angstrom**5

Maximum field normal   :      16.61 kcal/mol.Angstrom
Minimum field normal   :     -41.48 kcal/mol.Angstrom
Mean field             :         -0.45 kcal/mol.Angstrom
Field range           :      58.09 kcal/mol.Angstrom
Total field variance   :      63.25 (kcal/mol.Angstrom)**2
+ve field variance     :       6.65 (kcal/mol.Angstrom)**2
-ve field variance     :      66.54 (kcal/mol.Angstrom)**2
Field balance param.   :          0.08
Field skew             :          2.34
Field kurtosis         :          4.278
Integral F(N)          :       51.51      kcal.Angstrom/mol
Integral F(N +ve)      :      1332.      kcal.Angstrom/mol
Integral F(N -ve)      :     -1281.      kcal.Angstrom/mol
Integral |F(N)|        :      2613.      kcal.Angstrom/mol

<> Spherical-Harmonic Hybridization:

Shape hybrids          :
17.576151      1.110861      3.450758      2.848775      1.410632      1.602015
0.752068      0.688202      0.462321      0.389758      0.387721      0.318766
0.254783      0.212811      0.209028      0.200477

MEP hybrids            :
13.217889      4.729529      26.188350      18.489609      13.992226      10.724730
7.220981      8.124247      4.622006      4.919896      4.395234      3.580830
2.541562      2.738907      1.629285      2.033261      1.920217      1.871438
1.474428      1.035562      1.017963

IE(l) hybrids          :
1698.1619      90.2420      81.3581      62.4739      59.3195      54.6003
53.6317      54.9183      40.0437      44.6963      25.5245      24.8963
19.8380      15.9993      18.4093      15.7391      14.4282      16.4144
14.7771      0.0000      0.0000

EA(l) hybrids          :
324.0302      6.1868      12.8590      25.9665      13.9365      16.5139
8.5527      9.0223      8.0923      7.5131      6.2313      5.3970
5.1376      7.5750      5.6376      4.4820      4.8386      3.7336
3.7363      3.6150      3.4503

Alpha(l) hybrids       :
1.01371172      0.05270984      0.03338628      0.02985029      0.02186603      0.01485299
0.01197223      0.00890841      0.00764054      0.00889521      0.00767265      0.00638097
0.00682204      0.00625396      0.00851990      0.00656432      0.00000000      0.00000000
0.00000000      0.00000000      0.00000000
```



Figure 9 continued

Field(N) hybrids :										
	2.6423	1.8631	13.5263	10.4406	10.4941	8.0651				
	6.0754	7.7827	5.4503	5.8168	5.6348	5.4245				
	4.7299	4.2978	2.4961	3.4376	3.8431	4.0455				
	3.3698	2.1922	1.9336							
<> Standard rotationally invariant fingerprint:(L. Mavridis, B. D. Hudson and D. W. Ritchie, J. Chem. Inf. Model., 2007, 47, 1787-1796.)										
	4.19239	1.05397	1.85762	1.68783	1.18770					
	1.26571	0.867219	3.63564	2.17475	5.11746					
	4.29995	3.74062	3.27486	2.68719	2.85031					
	2.14989	2.21808	2.09648	1.89231	1.59423					
	41.2088	9.49958	9.01987	7.90404	7.70191					
	7.38920	7.32337	18.0008	2.48732	3.58595					
	5.09574	3.73317	4.06373	2.92450	1.00683					
	0.229586	0.182719	0.172772	0.147872	0.121873					
	0.109418	1.62552	1.36495	3.67782	3.23120					
	3.23946	2.83992	2.46484	2.78974	2.33458					
	2.41180	2.37377	2.32906	2.17484						
<> Atomic surface properties:										
Atom	Area	MEP		IE(l)		EA(l)		mean	Field(N)	
		max	min	max	min	max	min	pol.	max	min
C	1	0.000								
O	2	0.073	-38.63	-41.63	535.57	533.00	-77.38	-79.30	0.260	-21.49 -27.99
C	3	3.380	-6.37	-46.93	578.39	474.06	-41.34	-89.55	0.307	-2.87 -30.99
C	4	1.259	-5.18	-15.61	573.38	498.15	-67.78	-90.31	0.322	-4.24 -8.83
C	5	0.699	-8.34	-13.91	573.56	531.88	-83.99	-92.50	0.320	-1.60 -7.33
C	6	0.000								
C	7	0.803	-9.81	-15.04	559.06	516.48	-71.63	-91.20	0.319	-2.88 -9.76
C	8	4.155	-1.37	-21.94	585.08	484.62	-48.40	-95.76	0.295	2.87 -18.92
N	9	4.183	-14.72	-32.41	535.62	452.64	-79.87	-104.94	0.279	-13.08 -41.48
C	10	10.329	-1.83	-27.19	633.65	532.84	-38.83	-86.77	0.284	3.35 -26.41
N	11	0.000								
N	12	1.441	-14.68	-33.29	537.97	472.79	-74.67	-98.55	0.268	-16.13 -37.17
C	13	6.263	-8.51	-26.50	636.94	512.98	-38.27	-82.53	0.286	-0.32 -26.69
N	14	0.000								
C	15	2.104	-7.01	-15.40	589.73	496.26	-64.25	-92.16	0.316	-3.39 -9.83
C	16	3.889	-11.80	-43.40	569.47	479.95	-40.62	-85.85	0.310	-4.76 -35.29
O	17	0.000								
C	18	0.000								
C	19	6.191	-17.34	-48.51	583.26	465.76	-46.51	-85.87	0.315	-6.05 -34.36
O	20	2.039	-31.11	-44.39	532.30	445.12	-75.33	-93.15	0.247	-13.38 -34.51
C	21	0.000								
H	22	32.635	13.06	-39.51	560.91	405.93	-82.22	-99.65	0.297	8.87 -25.66
H	23	21.584	14.07	-20.11	561.95	408.24	-83.73	-96.09	0.294	9.65 -28.49
H	24	24.164	14.04	-30.22	567.47	406.95	-68.87	-95.30	0.293	9.71 -17.83
H	25	6.870	11.16	-5.33	527.22	425.98	-83.09	-95.66	0.288	7.63 -5.92
H	26	18.506	7.90	-7.44	535.28	400.17	-87.74	-100.67	0.300	5.64 -5.92
H	27	17.885	7.35	-20.61	543.51	399.68	-68.57	-100.82	0.303	4.52 -21.74
H	28	26.817	8.86	-27.87	536.72	413.57	-60.29	-103.43	0.284	5.88 -32.28
H	29	33.031	16.80	-28.67	671.23	474.66	-72.02	-107.86	0.248	16.61 -32.31
H	30	32.854	16.44	-28.19	669.05	476.80	-72.56	-107.92	0.241	16.30 -32.17
H	31	33.925	11.44	-28.94	657.13	471.46	-67.51	-108.53	0.244	12.43 -35.12
H	32	9.936	11.30	-27.10	641.75	472.59	-73.11	-99.95	0.258	12.66 -34.70
H	33	5.297	9.88	-13.59	504.99	434.61	-73.78	-96.58	0.295	7.73 -9.33
H	34	29.288	11.33	-40.38	564.99	406.94	-80.27	-99.69	0.291	7.86 -33.46
H	35	23.557	11.45	-33.33	567.57	407.56	-72.90	-95.67	0.293	8.06 -23.22
H	36	23.013	11.40	-13.11	563.83	407.55	-81.57	-99.02	0.295	8.02 -12.33
H	37	33.967	4.59	-38.60	554.53	394.86	-87.16	-108.54	0.295	4.15 -30.30
H	38	18.879	4.19	-40.99	558.76	396.53	-84.56	-108.22	0.298	3.76 -25.13
H	39	26.447	3.97	-34.03	550.40	391.03	-86.19	-107.60	0.296	3.63 -26.93
Total 465.463										
<> ParaSurf used 6.86 seconds CPU time										



ParaSurf10™ first fits the calculated shrink-wrap surface at full resolution for each of the local properties. It lists the root-mean-square deviations (RMSDs) for the surface points as a function of the order of the spherical-harmonic expansion, first for the geometry of the surface and then for each of the five local properties. The RMSD values give an idea of how well each order of the spherical-harmonic expansion fits the calculated shrink-wrap surface or the relevant property. The highest order used by ParaSurf™ is 15 for the surface itself and 20 for each property.

The descriptor table is then printed. For molecules with no surface areas with positive EA_L , $\sigma_{EA_L}^2$ is set to zero. The descriptors are those described in Table 1.

The spherical-harmonic hybridization coefficients are then listed for the shape and the five local properties. The coefficients are listed by increasing l starting from zero. The standard rotationally invariant fingerprint (RIF) [32] is printed. Note that the individual RIF-values correspond to the square roots of the hybridization coefficients from the tables above and that the RIF definition has been expanded to include hybridization coefficients of the field normal to the surface (the last 13 elements).

The table of atomic surface properties is derived by first finding the atom that contributes most (according to a Coulson analysis) to the electron density for each surface point. The point is then assigned to this atom and the maxima and minima in the MEP, IE_L , EA_L and F_N as well as the mean local polarisability for the points assigned to each atom are calculated. Note that, because of the fitting procedure, the values reported in this table may contain spurious ones if the fitted surface comes particularly close to an atom (or does not approach it). This situation is generally recognisable from the RMSD values printed for the fit. The surface used to calculate the descriptors and atomic-surface properties is the fitted spherical-harmonic surface of order 15.



3.4.2 For a marching-cube surface

Figure 10 shows the output for a calculation using the options **surf=cube** for trimethoprim.

```
<> ParaSurf'10 Academic Version
<> Copyright (c) 2006,2007,2008,2009,2010 Friedrich-Alexander-Universitaet
    Erlangen-Nuernberg and Cepos InSilico Ltd.
    All rights reserved.

<> Input = trimethoprim.sdf

<> Program options :

    Using marching-cube isodensity surface
    Surface fitting turned off
    Using an isodensity surface contour
    Isodensity value = 0.3000E-03 electrons/Angstrom**3
    Triangulation mesh = 0.20 Angstrom
    Using multipole electrostatics

<> AM1 calculation for Trimethoprim
<> Number of triangles = 15024
<> Number of unique points : 7517

<> Property ranges:
Density : 0.2879E-03 to 0.3111E-03
IE(1) : 392.33 to 654.93
EA(1) : -109.82 to -28.98
MEP : -69.91 to 24.82
Alpha(1) : 0.2288 to 0.3302
Field(N) : -106.67 to 72.35

<> Descriptors :

Dipole moment : 1.2486 Debye
Dipolar density : 0.3160E-02 Debye.Angstrom**-3
Molecular pol. : 31.2348 Angstrom**3
Molecular weight : 290.32
Globularity : 0.7042
Total surface area : 369.81 Angstrom**2
Molecular volume : 395.14 Angstrom**3

Most positive MEP : 24.82 kcal/mol
Most negative MEP : -69.91 kcal/mol
Mean +ve MEP : 9.04 kcal/mol
Mean -ve MEP : -18.73 kcal/mol
Mean MEP : -4.94 kcal/mol
MEP range : 94.73 kcal/mol
MEP +ve Variance : 31.62 (kcal/mol)**2
MEP -ve Variance : 240.01 (kcal/mol)**2
MEP total variance : 271.63 (kcal/mol)**2
MEP balance parameter: 0.1028
MEP balance*variance : 27.9357 kcal/mol
MEP skewness : -1.0235
MEP kurtosis : 0.6089
Integral MEP : -1674.03 kcal.Angstrom**2/mol

Maximum IE(1) : 654.93 kcal/mol
Minimum IE(1) : 392.33 kcal/mol
Mean IE(1) : 486.25 kcal/mol
IE(1) range : 262.60 kcal/mol
IE(1) variance : 3586.57 (kcal/mol)**2
IE(1) skewness : 0.4205
IE(1) kurtosis : -0.7628
Integral IE(1) : 7764.79 eV.Angstrom**2
```

Figure 10 ParaSurf™ output for trimethoprim using a marching-cube surface



Figure 10 continued

```
Maximum EA(l)      :      -28.98 kcal/mol
Minimum EA(l)      :     -109.82 kcal/mol
Mean +ve EA(l)     :         0.00 kcal/mol
Mean -ve EA(l)     :     -89.08 kcal/mol
Mean EA(l)         :     -89.08 kcal/mol
EA(l) range        :         80.83 kcal/mol
EA(l) +ve variance :         0.00 (kcal/mol)**2
EA(l) -ve variance :        276.48 (kcal/mol)**2
EA(l) total variance :        276.48 (kcal/mol)**2
EA(l) skewness     :         1.4612
EA(l) kurtosis     :         1.5705
Integral EA(l)      :    -1438.96 eV.Angstrom**2
EA(l) balance param. :         0.0000
Fraction pos. EA(l) :         1.0000 ( = 369.81 Angstrom**2)

Max. local Eneg.    :        290.20 kcal/mol
Min. local Eneg.    :        143.77 kcal/mol
Mean local Eneg.    :        198.59 kcal/mol
Local Eneg. range   :        146.43 kcal/mol
Local Eneg. variance :       1206.49 (kcal/mol)**2
Local Eneg. skewness :          0.52
Local Eneg. kurtosis :        -0.78
Integral local Eneg. :    3162.92 eV.Angstrom**2

Max. local hardness :        371.14 kcal/mol
Min. local hardness :        247.89 kcal/mol
Mean local hardness :        287.67 kcal/mol
Local hard. range   :        123.25 kcal/mol
Local hard. variance :       725.03 (kcal/mol)**2
Local hard. skewness :          0.45
Local hard. kurtosis :        -0.66
Integral local Hard. :    4601.87 eV.Angstrom**2

Maximum alpha(l)    :        0.3302 Angstrom**3
Minimum alpha(l)    :        0.2288 Angstrom**3
Mean alpha(l)       :        0.2830 Angstrom**3
Alpha(l) range      :        0.1014 Angstrom**3
Variance in alpha(l) :    0.4904E-03 Angstrom**6
Alpha(l) skewness   :        -0.8023
Alpha(l) kurtosis   :        -0.3815
Integral Alpha(l)    :    104.504 Angstrom**5

Maximum field normal :        72.35 kcal/mol.Angstrom
Minimum field normal :   -106.67 kcal/mol.Angstrom
Mean field           :        -0.35 kcal/mol.Angstrom
Field range          :        179.02 kcal/mol.Angstrom
Total field variance :    398.24 (kcal/mol.Angstrom)**2
+ve field variance   :        35.02 (kcal/mol.Angstrom)**2
-ve field variance   :    585.33 (kcal/mol.Angstrom)**2
Field balance param. :          0.05
Field skew           :         2.74
Field kurtosis       :        6.578
Integral F(N)        :        69.53 kcal.Angstrom/mol
Integral F(N +ve)    :        2364. kcal.Angstrom/mol
Integral F(N -ve)    :       -2295. kcal.Angstrom/mol
Integral |F(N)|      :        4659. kcal.Angstrom/mol
```



Figure 10 continued

<> Atomic surface properties:												
Atom		Area	MEP		IE (l)		EA (l)		mean	Field (N)		
			max	min	max	min	max	min	pol.	max	min	
C	1	0.255	-24.20	-46.87	569.21	546.16	-81.57	-92.67	0.268	-7.45	-44.94	
O	2	3.663	-15.51	-69.27	594.82	456.70	-63.60	-81.53	0.269	23.64	-106.67	
C	3	6.489	-7.34	-64.83	642.96	499.79	-30.30	-99.31	0.304	1.34	-84.49	
C	4	2.172	-3.32	-19.78	631.80	493.54	-40.13	-100.66	0.316	-2.82	-20.02	
C	5	1.621	-2.14	-18.29	633.57	546.99	-53.30	-100.26	0.313	3.29	-11.55	
C	6	0.000										
C	7	2.050	-4.01	-22.32	605.59	512.50	-50.08	-91.07	0.317	0.14	-26.55	
C	8	5.598	4.10	-28.25	638.27	488.28	-35.86	-88.17	0.288	9.86	-29.36	
N	9	6.696	-19.85	-58.79	571.47	417.77	-54.64	-103.25	0.260	-14.68	-93.22	
C	10	9.420	-0.80	-46.25	654.93	543.27	-41.16	-81.01	0.280	13.28	-61.33	
N	11	0.536	-46.54	-53.02	615.74	593.92	-60.29	-79.02	0.276	-60.55	-81.55	
N	12	6.115	-16.63	-55.65	571.35	417.33	-51.08	-98.91	0.247	-15.21	-91.12	
C	13	7.551	-10.11	-44.06	644.97	527.96	-37.44	-82.79	0.284	10.94	-58.77	
N	14	0.713	-41.89	-57.14	618.69	590.74	-64.83	-81.20	0.287	-40.63	-95.16	
C	15	4.193	-8.08	-22.32	640.33	494.10	-30.85	-100.79	0.314	0.50	-22.28	
C	16	5.831	-15.48	-61.39	641.08	507.48	-28.98	-94.37	0.307	0.56	-75.79	
O	17	1.251	-18.69	-69.91	566.92	464.96	-64.24	-87.63	0.252	-11.13	-104.49	
C	18	0.290	-16.74	-56.74	573.31	531.64	-74.36	-94.92	0.267	-7.21	-55.54	
C	19	5.583	-15.50	-60.30	617.67	492.45	-39.17	-96.41	0.314	-0.58	-50.34	
O	20	3.968	-31.04	-63.58	581.05	438.59	-68.75	-94.46	0.265	-3.30	-76.06	
C	21	0.545	-26.28	-54.85	563.65	530.77	-90.80	-106.12	0.269	-0.65	-62.76	
H	22	20.849	22.24	-44.01	561.09	407.48	-83.43	-97.55	0.297	19.34	-39.51	
H	23	16.018	22.20	-47.62	566.04	408.23	-70.44	-97.22	0.293	19.83	-89.24	
H	24	16.231	22.21	-45.82	567.70	407.67	-66.85	-96.73	0.291	19.67	-51.22	
H	25	7.150	16.35	-8.07	537.75	430.17	-70.34	-97.50	0.287	14.85	-11.83	
H	26	13.485	13.09	-5.84	570.04	401.19	-85.82	-100.62	0.299	9.80	-10.05	
H	27	13.124	11.65	-38.34	610.80	400.49	-74.20	-100.86	0.301	13.51	-51.85	
H	28	17.543	13.34	-29.83	533.67	415.37	-53.69	-100.29	0.282	18.51	-31.14	
H	29	20.142	24.45	-44.08	639.93	488.10	-72.39	-107.70	0.247	33.70	-70.29	
H	30	20.327	24.82	-48.23	644.32	488.09	-70.46	-107.77	0.241	29.18	-71.86	
H	31	20.033	22.74	-51.84	642.48	483.81	-67.40	-108.33	0.245	32.52	-81.11	
H	32	10.797	22.33	-49.42	644.51	478.85	-79.21	-102.14	0.258	26.11	-87.40	
H	33	7.985	15.01	-21.60	523.96	429.37	-65.63	-98.60	0.294	10.93	-18.28	
H	34	20.354	17.51	-31.88	560.04	408.30	-87.92	-99.03	0.290	18.83	-16.57	
H	35	16.207	17.81	-47.27	565.70	408.17	-66.88	-96.52	0.291	15.15	-40.08	
H	36	16.208	17.77	-37.50	557.34	408.12	-69.21	-96.79	0.294	15.10	-55.02	
H	37	20.706	8.06	-41.68	545.25	396.03	-95.88	-109.82	0.294	10.46	-27.25	
H	38	16.212	8.06	-54.60	594.17	394.19	-82.67	-109.53	0.296	72.35	-73.80	
H	39	18.662	7.90	-40.12	586.35	392.33	-74.88	-109.41	0.295	26.55	-32.47	
Total		366.571										
<> ParaSurf used			5.75 seconds CPU time									

The table of RMSD values is no longer printed and the range of the electron-density values for the surface points (a test for the quality of the surface) is closer to the target isodensity value (in this case $0.0003 \text{ e}^- \text{\AA}^{-3}$) than for the fitted surface. The internal precision used by the program is $\pm 2\%$ of the target isodensity value. The values of the descriptors and the atomic-surface properties are more consistent using the marching-cube surface and are recommended for QSPR and surface-integral applications.

3.4.3 For a job with Shannon entropy*

3.4.4 For a job with autocorrelation similarity*

* Only available in the full version.



3.5 ParaSurf™ SDF-output

The SDF output file (a fixed-format file) contains additional blocks with the information generated by ParaSurf™. These are:

<ParaSurf OPTIONS>

The ParaSurf™ OPTIONS block consists of one line giving the options used in the ParaSurf™ calculation. These are:

<surface> <fit> <electrostatic model> <isodensity level> (a4,2x,a4,2x,a5,2x,f8.3)

Where the individual variables can be:

<surface>	WRAP	Shrink-wrap surface
	CUBE	Marching-cube surface
<fit>	NONE	No fitting, unsmoothed marching-cube surface
	ISO	Marching-cube surface corrected to $\pm 2\%$ of the preset isodensity value
	SPHH	Spherical-harmonic surface fit
<electrostatic model>	NAOPC	NAO-PC electrostatics
	MULTI	Multipole electrostatics
<isodensity level>	n.nn	The target isodensity value in $e^{-\text{\AA}^{-3}}$
<solvent probe radius>		The radius of the solvent probe used to calculate the SES or SAS
<triangulation mesh>		The mesh size used to triangulate the Surface

<MOLECULAR_CENTERS>

The molecular centres block appears only for calculations that use spherical harmonic fits. It includes two lines of the form:

```
"Spherical harmonic center = ", 3f12.6
"Center of gravity          = ", 3f12.6
```

These blocks give the x, y and z coordinates of the centre of the molecule used for the spherical-harmonic fit and the centre of gravity, respectively. These two centres are usually identical, but may be different if the centre of gravity lies outside the molecule (e.g. for U-shaped molecules).

<SPHERICAL_HARMONIC_.....>

The spherical harmonic fits are described in <SPHERICAL_HARMONIC_....> blocks. These blocks all have the same format and vary only in the property described. Each block has the form:

The spherical harmonic fits are described in <SPHERICAL_HARMONIC_.....> blocks. These blocks all have the same format and vary only in the property described. Each block has the form:



Order = nn	("Order = ", i4)
$l(C_l^m)m = -l \text{ to } l$	(I5, 10f8.4/5x, 10f8.4/5x, 10f8.4/5x, 10f8.4) (One set of coefficients each for $l = 1$ to 15)
RMSDs: $l, \text{RMSD}^1, \text{RMSD}^2$	("RMSDs:") (i8, 2f12.8) (One line for each l for $l = 1$ to 15, where RMSD^1 is the area-weighted RMSD and RMSD^2 the simple RMSD)

There are six such blocks, indicated by the tags:

<SPHERICAL_HARMONIC_SURFACE>	The fitted molecular surface (radial distances) in Ångstrom
<SPHERICAL_HARMONIC_MEP>	The MEP values at the spherical-harmonic surface ($l = 20$) in kcal mol ⁻¹
<SPHERICAL_HARMONIC_IE(l)>	The IE _L values at the spherical-harmonic surface ($l = 20$) in kcal mol ⁻¹
<SPHERICAL_HARMONIC_EA(l)>	The EA _L values at the spherical-harmonic surface ($l = 20$) in kcal mol ⁻¹
<SPHERICAL_HARMONIC_ALPHA(l)>	The α _L values at the spherical-harmonic surface ($l = 20$) in kcal mol ⁻¹
<SPHERICAL_HARMONIC_FIELD(N)>	The FN values at the spherical-harmonic surface ($l = 20$) in kcal mol ⁻¹ Å ⁻¹

<ParaSurf Descriptors>

The ParaSurf™ descriptors block lists the calculated descriptors in the following groups:

Molecular:	μ, μ _D , α, MW, G, A, VOL ("Molecular ", 5f10.4, 2f10.2)
MEP:	$V_{\max}, V_{\min}, \bar{V}_+, \bar{V}_-, \bar{V}, \Delta V, \sigma_+^2, \sigma_-^2, \sigma_{\text{Tot}}^2, \nu, \sigma_{\text{tot}}^2 \nu, \gamma_1^V, \gamma_2^V, \int_V$ ("MEP ", 7f10.2/10x, f10.2, 5f10.4, 2x, g12.6)
IE(l):	$IE_L^{\max}, IE_L^{\min}, \bar{IE}_L, \Delta IE_L, \sigma_{IE}^2, \gamma_1^{IE}, \gamma_2^{IE}, \int_{IE}$ ("IE (l) ", 5f10.2, 2f10.4/12x, g12.6)
EA(l):	$EA_L^{\max}, EA_L^{\min}, \bar{EA}_L, \Delta EA_L, \sigma_{EA}^2, \gamma_1^{EA}, \gamma_2^{EA}, \int_{EA}$ ("EA (l) ", 7f10.2/2f10.2, 2f10.4, f10.2, 2f10.4/12x, g12.6)
Eneg(l):	$\chi_L^{\max}, \chi_L^{\min}, \bar{\chi}_L, \Delta \chi_L, \sigma_{\chi}^2, \gamma_1^{\chi}, \gamma_2^{\chi}, \int_{\chi}$ ("Eneg (l) ", 5f10.2, 2f10.4/12x, g12.6)
Hard(l):	$\eta_L^{\max}, \eta_L^{\min}, \bar{\eta}_L, \Delta \eta_L, \sigma_{\eta}^2, \gamma_1^{\eta}, \gamma_2^{\eta}, \int_{\eta}$ ("Hard (l) ", 5f10.2, 2f10.4/12x, g12.6)
Alpha(l):	$\alpha_L^{\max}, \alpha_L^{\min}, \bar{\alpha}_L, \Delta \alpha_L, \sigma_{\alpha}^2, \gamma_1^{\alpha}, \gamma_2^{\alpha}, \int_{\alpha}$ ("Alpha (l) ", 5f10.2, 2f10.4/12x, g12.6)
F_N	$F_N^{\max}, F_N^{\min}, \bar{F}_N, \sigma_F^2, \sigma_{F+}^2, \sigma_{F-}^2, \nu_F, \gamma_1^{F_N}, \gamma_2^{F_N}, \int_{F_N}, \int_{F_N}^+, \int_{F_N}^-, \int_{F_N}$ ("Field desc", 7f10.4/" ", 6f10.4)

For calculations using a spherical-harmonic fit, the hybridization coefficients are printed to the .sdf file as follows (tag line followed by as many lines with the coefficients as necessary):

<SHAPE HYBRIDS>

(15 coefficients, 6f12.6)

<MEP HYBRIDS>

(20 coefficients, 6f12.6)

<IE(l) HYBRIDS>

(20 coefficients, 6f12.2)



<EA(L) HYBRIDS>

(20 coefficients, 6f12.2)

<ALPHA(L) HYBRIDS>

(20 coefficients, 6f12.8)

<FIELD(N) HYBRIDS>

(20 coefficients, 6f12.4)

The hybridization coefficients are listed in order of increasing l from zero, exactly as in the output file.

The atomic surface properties are listed in the atomic order according to the following headings (tag line followed by as many lines with the surface properties as necessary):

<ATOMIC SURFACE AREAS>

Areas (10f8.4)

<ATOMIC SURFACE MEP MAXIMA>

MEP maxima (10f8.2)

<ATOMIC SURFACE MEP MINIMA>

MEP minima (10f8.2)

<ATOMIC SURFACE IE(L) MAXIMA>

IE(l) maxima (10f8.2)

<ATOMIC SURFACE IE(L) MINIMA>

IE(l) minima (10f8.2)

<ATOMIC SURFACE EA(L) MAXIMA>

EA(l) maxima (10f8.2)

<ATOMIC SURFACE EA(L) MINIMA>

EA(l) minima (10f8.2)

<ATOMIC SURFACE MEAN POL>

Mean pol. (10f8.4)

<ATOMIC SURFACE FIELD(N) MAXIMA>

FN maxima (10f8.2)

<ATOMIC SURFACE FIELD(N) MINIMA>

FN minima (10f8.2)

<STANDARD RIF>

The rotationally invariant fingerprint [32] is printed as a list of 54 floating point numbers (5g12.6). The first 41 are those defined in reference [32] and the last 13 are the square roots of the hybridization coefficients for the normal field from $l=0-12$.

Hard(l)	$\eta_L^{\max}, \eta_L^{\min}, \overline{\eta_L}, \Delta\eta_L, \sigma_\eta^2, \gamma_1^\eta, \gamma_2^\eta, \int_\eta$ ("Hard(l) ", 5f10.2, 2f10.4/12x, g12.6)
Alpha(l)	$\alpha_L^{\max}, \alpha_L^{\min}, \overline{\alpha_L}, \Delta\alpha_L, \sigma_\alpha^2, \gamma_1^\alpha, \gamma_2^\alpha, \int_\alpha$ ("Alpha(l) ", 5f10.2, 2f10.4/12x, g12.6)
F _N	$F_N^{\max}, F_N^{\min}, \Delta F_N, \overline{F_N}, \sigma_F^2, \sigma_{F+}^2, \sigma_{F-}^2, \nu_F, \gamma_1^{F_N}, \gamma_2^{F_N}, \int_{F_N}, \int_{F_N}^+, \int_{F_N}^-$ ("Field desc", 7f10.4/" ", 6f10.4)

3.5.1 Optional blocks in the SDF-output file*

3.6 The surface (.psf) file*

3.7 Anonymous SD (.asd) files*

3.7.1 Optional blocks*

* Only available in the full version.



3.8 Grid calculations with ParaSurf™

3.8.1 User-specified Grid

The command

```
parasurf <filename> grid=grid.dat
```

instructs ParaSurf™ to read a set of Cartesian coordinates from the file grid.dat and to calculate the four local properties (MEP, IEL, EAL, α L). The format of the file grid.dat (which must be in the same directory as the input) is one line per atom containing the x, y and z coordinates in free format, comma-separated, maximum line length 80. For instance, the following grid file:

```
-3.79480 , -7.06030 , 10.37150
-3.79480 , -5.06030 , -7.62850
-3.79480 , -5.06030 , -5.62850
-3.79480 , -5.06030 , -3.62850
-3.79480 , -5.06030 , 0.37150
-3.79480 , -5.06030 , 2.37150
-3.79480 , -5.06030 , 4.37150
-3.79480 , -5.06030 , 6.37150
-3.79480 , -5.06030 , 8.37150
-3.79480 , -5.06030 , 10.37150
-3.79480 , -3.06030 , -7.62850
-3.79480 , -3.06030 , -5.62850
-3.79480 , -3.06030 , 0.37150
-3.79480 , -3.06030 , 2.37150
```

Figure 11 Sample grid file

gives the output shown in Figure 12.



```
<> ParaSurf'10 Academic Version
<> Copyright (c) 2006,2007,2008,2009,2010 Friedrich-Alexander-Universitaet
    Erlangen-Nuernberg and cepos insilico Ltd.
    All rights reserved.

<> Input = trimethoprim.sdf
<> Program options :
    Calculating local properties using grid file grid.dat
    using multipole electrostatics

<> AM1 calculation for Trimethoprim

      x      y      z      density      MEP      IEL      EA      Pol      Ene      Hard      dv/dx      dv/dy      dv/dz
-3.79480 -7.06030 10.37150 0.1936E-21 0.30 492.06 -106.39 0.2415 192.84 293.22 -0.0097 -0.0048 0.0511
-3.79480 -5.06030 -7.62850 0.8572E-10 -2.55 473.35 -93.89 0.2934 189.73 283.62 1.4171 0.2751 0.3492
-3.79480 -5.06030 -5.62850 0.3544E-06 -2.70 486.70 -94.43 0.2953 196.13 290.57 3.7559 0.4948 0.5274
-3.79480 -5.06030 -3.62850 0.1630E-02 4.92 443.81 -92.79 0.2976 175.51 268.30 10.4560 0.2210 -12.2945
-3.79480 -5.06030 0.37150 0.1920E-04 2.71 466.84 -92.68 0.2955 187.08 279.76 4.3010 -1.1537 0.9171
-3.79480 -5.06030 2.37150 0.9497E-08 2.03 430.60 -91.71 0.2958 169.44 261.15 0.6472 0.7808 0.2089
-3.79480 -5.06030 4.37150 0.4718E-11 1.44 481.86 -96.44 0.2656 192.71 289.15 0.0148 0.3732 0.3411
-3.79480 -5.06030 6.37150 0.2516E-13 0.83 499.66 -104.66 0.2420 197.50 302.16 -0.0351 -0.0828 0.2492
-3.79480 -5.06030 8.37150 0.4712E-16 0.46 491.12 -106.67 0.2413 192.22 298.89 -0.0280 0.0126 0.1257
-3.79480 -5.06030 10.37150 0.3452E-19 0.29 487.25 -107.47 0.2412 189.89 297.36 -0.0236 0.0241 0.0578
-3.79480 -5.06030 -7.62850 0.4749E-09 -2.70 418.66 -92.37 0.2916 163.14 255.52 1.6383 -0.1541 0.4102
-3.79480 -3.06030 -5.62850 0.4085E-05 -2.69 415.34 -92.18 0.2915 161.58 253.76 4.6534 -0.4024 -1.0341
-3.79480 -3.06030 0.37150 0.4907E-03 8.27 409.38 -91.04 0.2956 159.17 250.21 3.9236 -2.7009 5.0984
-3.79480 -3.06030 2.37150 0.1385E-06 4.56 508.36 -90.91 0.2634 208.73 299.64 -1.2145 -1.6657 1.0536

<> ParaSurf used      0.03 seconds CPU time
```

Figure 12 Sample grid output file

The name and the extension (if any) of the grid file are free. Only the output file is written. The units of the local properties are those used in the normal output (i.e. V, IEL, and EAL in kcal mol⁻¹, α L in Ångström³).



3.8.2 Automatic grids

ParaSurf™ can generate grids automatically for lead compounds in ComFA®-like procedures. The **grid=auto** option generates a grid around the molecule (with a 4 Å margin around the positions of the atoms in each direction) and includes all points for which the electron density is lower than 10^{-2} (i.e. for points outside the molecule). The spacing of the grid is set to a default value of 1.0 Å, but can be set to any value up to a maximum of 2.0 Å by the command-line argument **lattice=*n.n***, which sets the lattice spacing to *n.n* Å. The grid thus generated is output (with the values of the local properties analogously to a calculation that uses an predefined grid and can be used for other molecules that have been aligned with the lead).

3.9 The SIM file format*

3.10 Output tables*

3.11 Binned SIM descriptor tables*

3.12 Autocorrelation similarity tables*

3.13 Shared files

The Vhamil.par and SIM files are accessed in shared, read-only mode so that multiple ParaSurf™ jobs can access the same files.

* Only available in the full version.



4 TIPS FOR USING PARASURF'10™

4.1 Choice of surface

ParaSurf™ was originally written to use isodensity surfaces. However, calculations that use a solvent-excluded surface are very much faster than their equivalents with isodensity surfaces and will usually give comparable results. Surface-integral models may benefit from using a solvent-excluded surface with a solvent radius of 0.5-1.0 Å as this appears to be the most relevant surface for many physical properties. Surfaces fitted to spherical-harmonic expansions require more CPU-time than marching-cube surfaces but are essential for fast numerical applications such as ParaFit™. Again, solvent-excluded shrink-wrap surfaces are faster to calculate than their isodensity equivalents.

4.2 ParaSurf™ and ParaFit™

ParaFit™ is Cepos InSilico's very fast shape-matching program that is based on spherical-harmonic expansions generated by ParaSurf™. ParaFit™ can be used to overlay molecules with a common scaffold by defining the centre to be used for generating the spherical-harmonic fit in ParaSurf™ in the input SDF-file (see 2.2)

4.3 QSAR using grids

As outlined in 3.8.2, ParaSurf™ can generate a grid for the lead molecule automatically that can then be used for a set of aligned (e.g. with ParaFit™) molecules for grid-based QSAR. This procedure has proven to be especially effective for test datasets, especially if the molecules are aligned to a common scaffold, as outlined in 4.2.

5 SUPPORT

5.1 Error reporting

Some of the routines in ParaSurf™ may detect error conditions that have not yet been encountered in our tests. In this case, an error message will be printed requesting that the input and output files be sent to the programming team at the above e-mail address. We realize that this will not always be possible for confidentiality reasons, but if the details can be sent, we will be able to treat the exception and improve the program.

5.2 CEPOS InSilico GmbH

Computer-Chemie-Centrum (CCC)
Nägelsbachstr. 25
91052 Erlangen
Germany



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