

# A Surface-Integral Model for $\log P_{ow}$ and a Local Hydrophobicity

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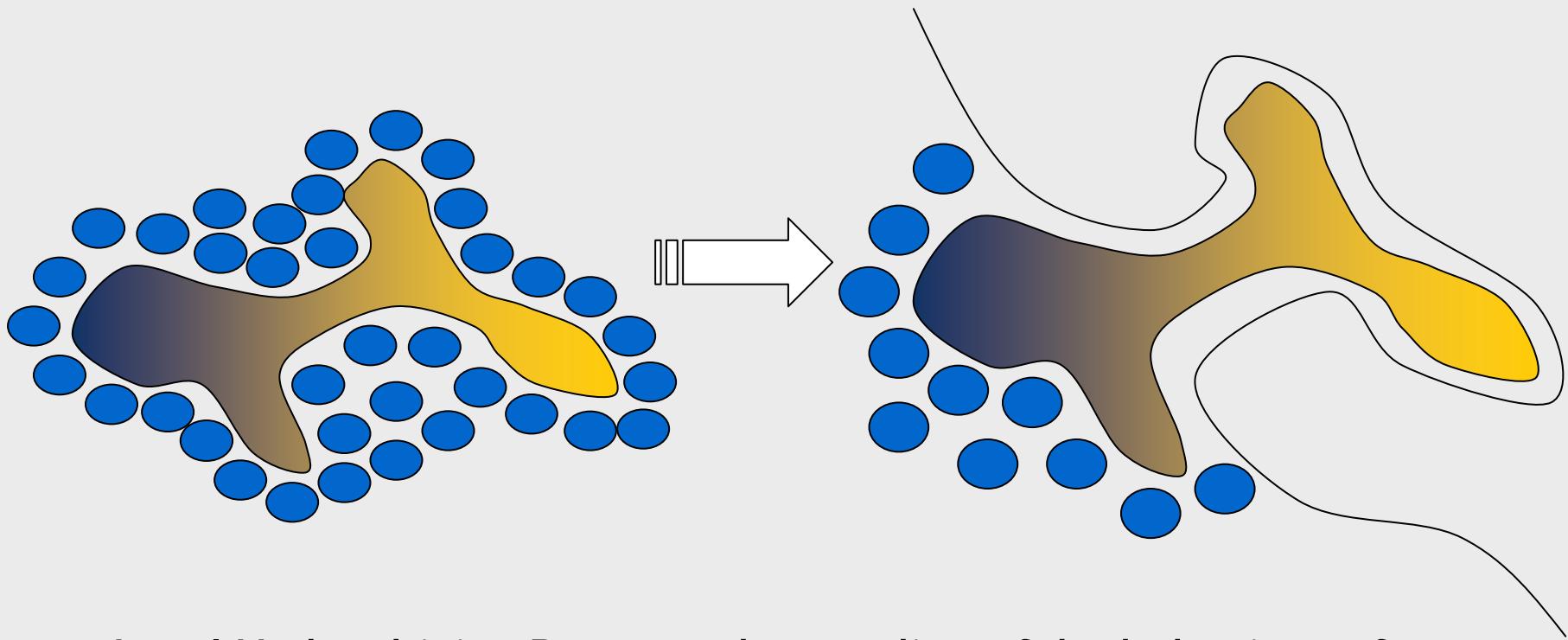
Boehringer  
Ingelheim

# Overview

- Introduction
- Calculation & Dataset
- Surface Integral Models
- Results
- Examples of Hydrophobic Surfaces
- Summary and Outlook

# Introduction - logP revisited

- $\log P_{ow}$  is an important descriptor in many QSAR approaches



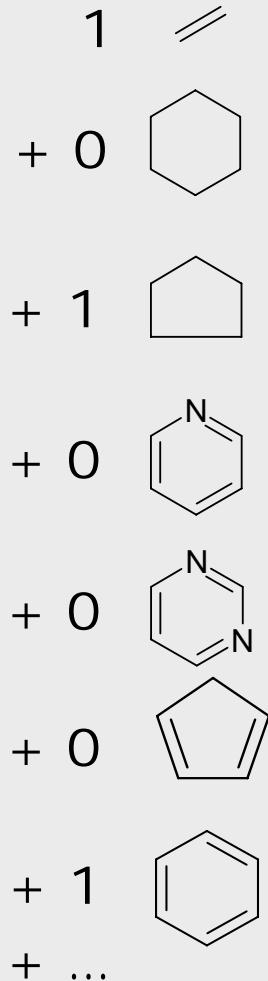
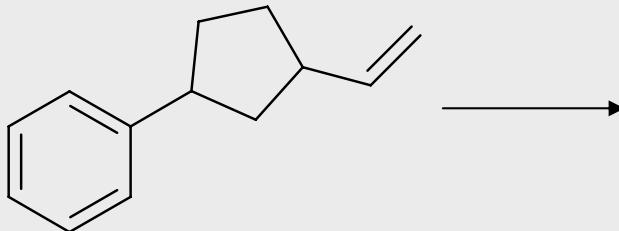
- Local Hydrophobicity: Better understanding of the behaviour of a molecule (visualization)
- Integral: Partition Coefficients

# Introduction

Locality:  
Substructural Fragments (e.g. ClogP)

Rekker<sup>[1]</sup>

Hansch & Leo<sup>[2]</sup>



[1] Nys, G.C.; Rekker, R.F. *Chim Ther* 1973, 8, 521

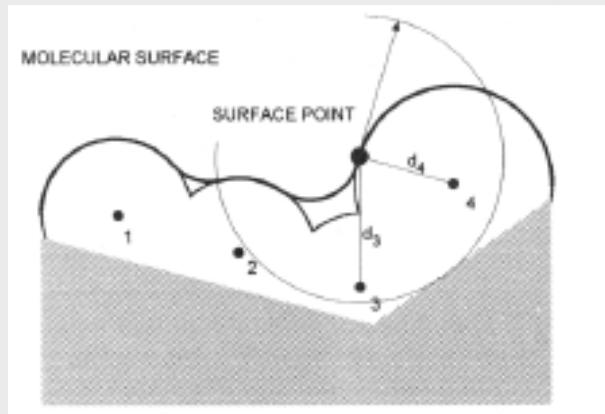
[2] Hansch, C.; Leo, A. *Substituent Constants for Correlation Analysis in Chemistry and Biology*, WILEY: New York, 1979

# Introduction

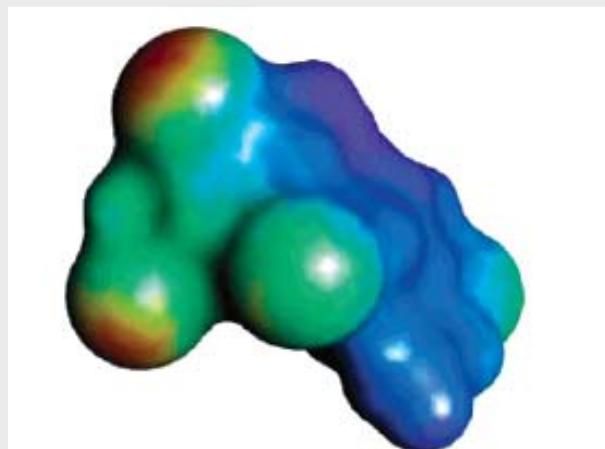
Map local hydrophobicity to the surface:

MolFESD approach 1 (1994, based on Ghose-Crippen Atom Types)<sup>[3]</sup>

MolFESD approach 2 (2003, based on MM-interactions with nonpolar probe on a GRID field)<sup>[4]</sup>



[3]



[4]

[3] Pixner, P.; Heiden, W.; Merx, H.; Moeckel, G.; Möller, A.; Brickmann, J. *J Chem Inf Comput Sci* 1994, 34, 1309-1319

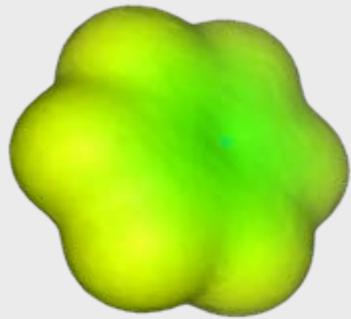
[4] Jäger, R.; Kast, S.M.; Brickmann, J. *J Chem Inf Comp Sci* 2003, 43, 237-247

# Calculation Procedure

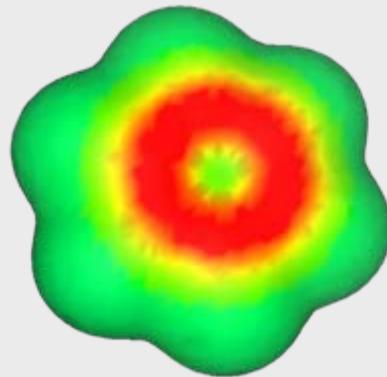
- CORINA (2D → 3D)
- VAMP (AM1, Geometry Optimization)
- ParaSurf (Surface @  $0.003\text{eA}^{-3}$ ; surface, surface property calculation)
  - MEP, IEL, EAL, HARD, POL, FN surfaces
- Calculation time around 10 to 20 seconds

# Calculated Local Properties

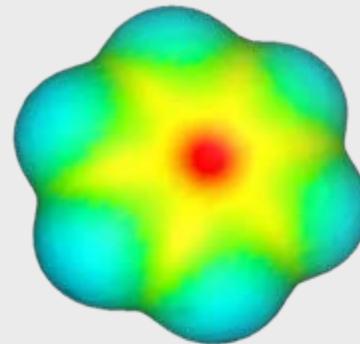
Our approach: Local Hydrophobicity as a function of QM Surface Properties



Molecular Electrostatic Potential<sup>[5]</sup>



Local Electron Affinity<sup>[7]</sup>



Local Ionization Energy<sup>[6]</sup>

+ Local Polarizability,<sup>[7]</sup> local Hardness,<sup>[7]</sup> Field normal to the surface<sup>[8]</sup>

[5] Politzer, P.; Truhlar, D.G. **1981**

[6] Sjoberg, P.; Murray, J.S.; Brinck, T.; Politzer, P. *Can J Chem* **1990**, *68*, 1440-1443

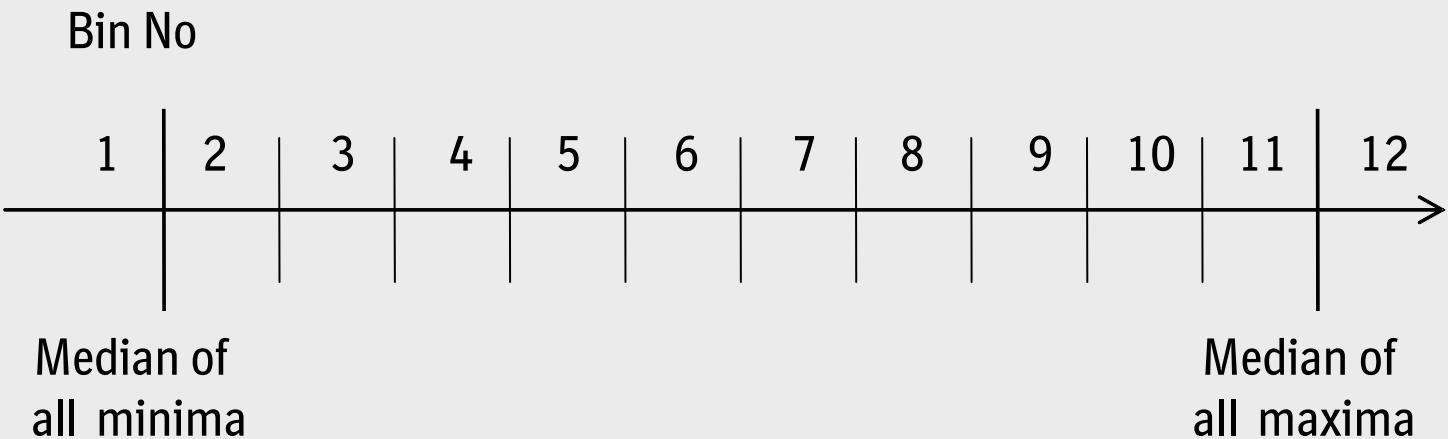
[7] Ehresmann, B.; de Groot, M.J.; Alex, A.; Clark, T. *J Chem Inf Comp Sci* **2004**, *43*, 658-668

[8] Clark, T. unpublished yet

# Dataset for Fitting

- Sangster Lab LOGKOW database
    - Dec. 2008: 37,783 values for 23,479 compounds
  - Quality check by Sangster Lab: Suggested good quality values:  
~11,500
  - Structural check, only H,C,N,O,F,P,S,Cl,Br,I compounds
  - Removal of zwitterions and permanently charged compounds
- 10,812 compounds, 12.5% randomly selected as validation set
- 768 Boehringer Ingelheim Compounds as additional validation set

# Surface Integral Models using: Binning

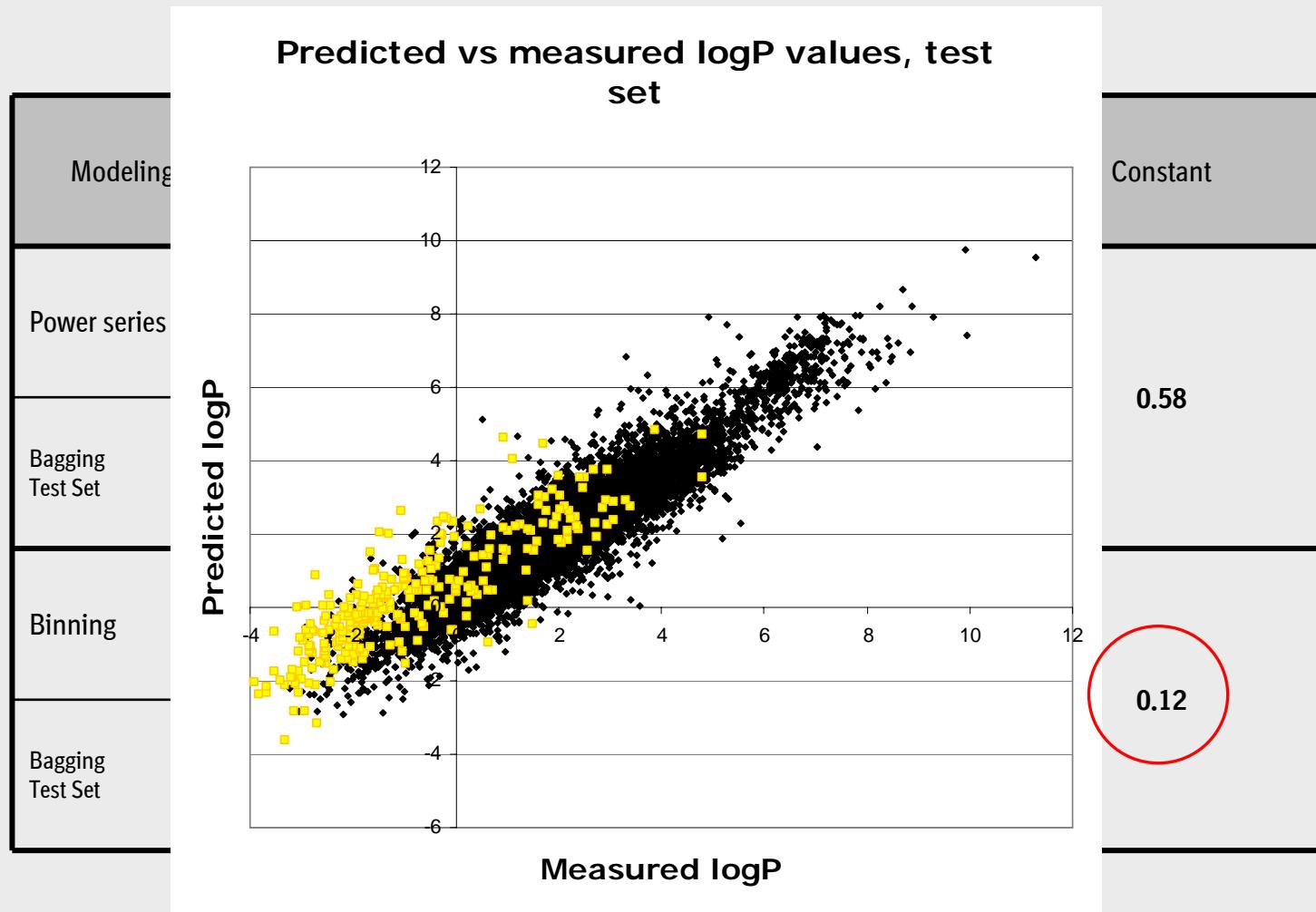


# Surface Integral Models: Fitting SIM Models

- Bagged stepwise multiple linear regression:
  - 9,462 collected and inspected  $\log P_{ow}$  values from James Sangster's LOGKOW database (<http://logkow.cisti.nrc.ca>)
  - 126 (Power Series) [9] or 252 (Binning) descriptors available
  - 50 bagging samples – 50 different models
  - 75% training set randomly chosen per sample
  - Test Set Performance evaluated after complete stepping
- Final regression contains 
$$\frac{\sum \text{coefficients}}{50}$$
- Most important descriptors:  $IE_L$ ,  $EA_L$ , local hardness and MEP

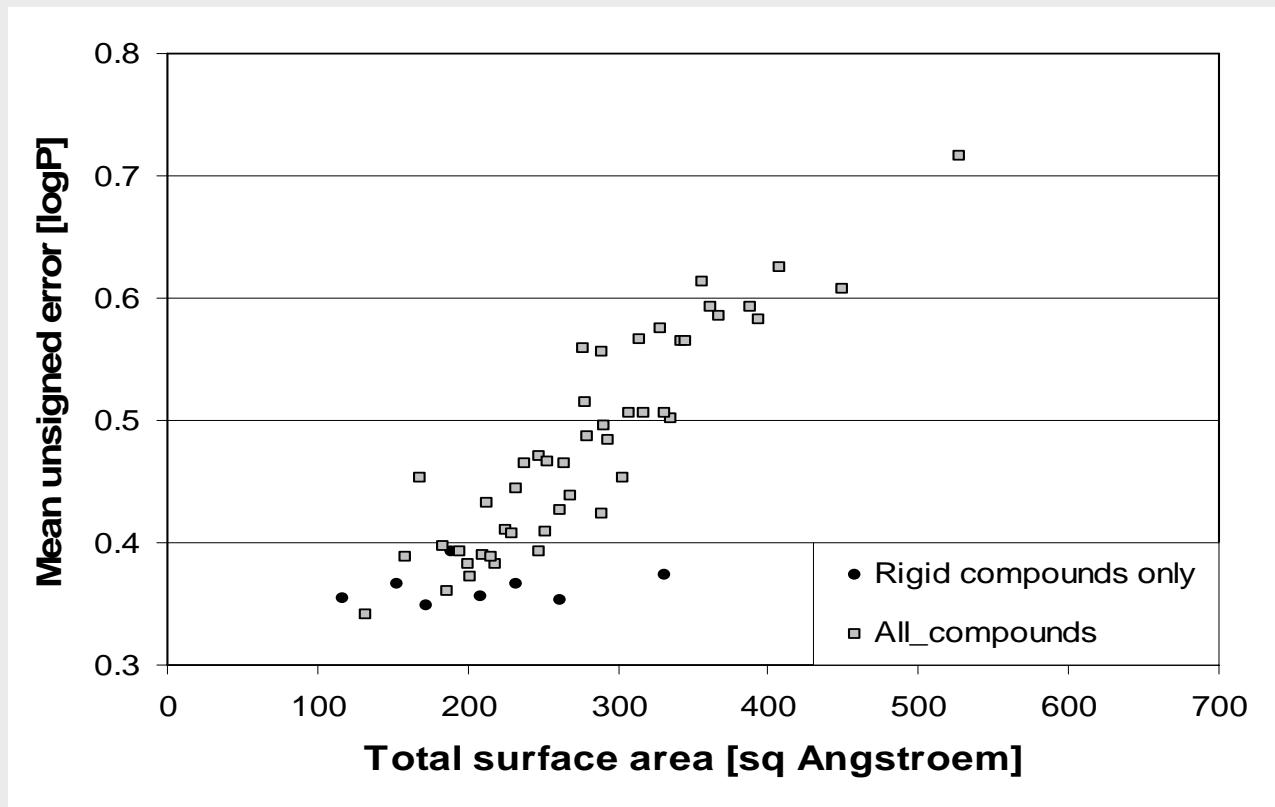
[9] Ehresmann, B.; de Groot, M.J.; Clark, T. *J Chem Inf Comp Sci* 2005, 45, 1053-1060

# Results: Integral of the local Hydrophobicity - logP

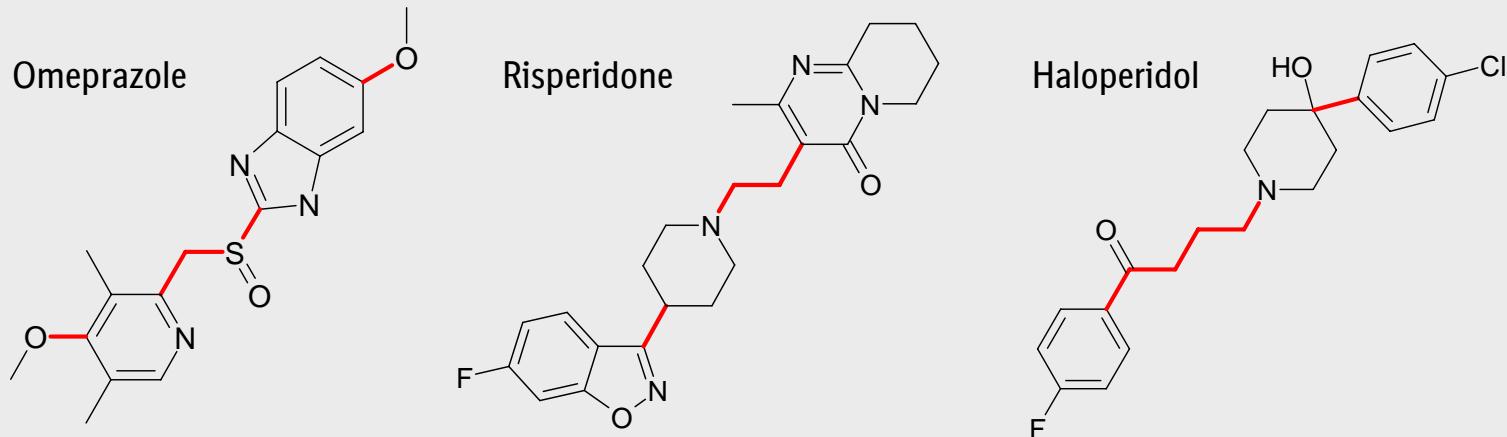


# Results: Error distribution - Rigid compounds only

Average mean unsigned error versus TotalSurface area,  
 comparison of compounds with and without rotatable  
 bonds



# Results: Improve prediction by conformational scanning



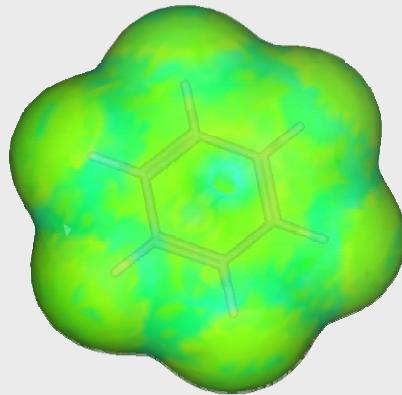
Compound	Experiment	Model	CORINA conformation	Boltzmann weighted	
				Gas-phase	SCRF
Omeprazole	2.23 2.38	Total Dataset	2.94	2.97	2.65
		Single Conf.	2.92	2.68	<b>2.51</b>
Risperidone	3.04	Total Dataset	2.48	3.20	3.14
		Single Conf.	2.78	3.17	<b>3.03</b>
Haloperidol	4.30	Total Dataset	4.52	4.23	4.19
		Single Conf.	4.17	4.20	4.18

# Results: Validation Set Performance

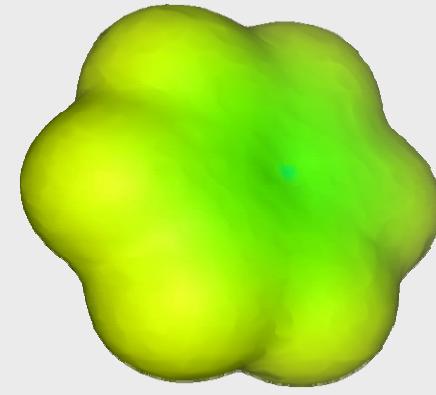
Model	Public Validation Set (1352 compounds)			Boehringer Ingelheim Validation Set (767 compounds)		
	MUE	RMSE	R <sup>2</sup>	MUE	RMSE	R <sup>2</sup>
SIM-logP	0.48	0.64	0.86	0.85	1.10	0.53
ACDlabs logP	0.26	0.45	0.94	1.03	1.36	0.47
ClogP	0.31	0.52	0.92	0.86	1.14	0.59
SlogP	0.53	0.68	0.85	0.92	1.19	0.51
logP_o/w	0.53	0.77	0.82	1.00	1.28	0.49
AlogP	0.62	0.86	0.79	0.97	1.24	0.50

# Results: Surface Integral Models for $\log P_{\text{OW}}$ for Benzene

Local  
Hydro-  
phobicity



Local  
Molecular  
Electrostatic  
Potential  
(MEP)

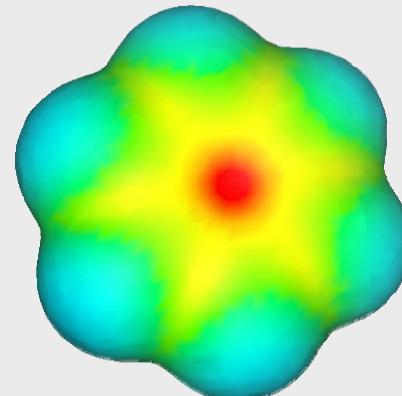


Benzene

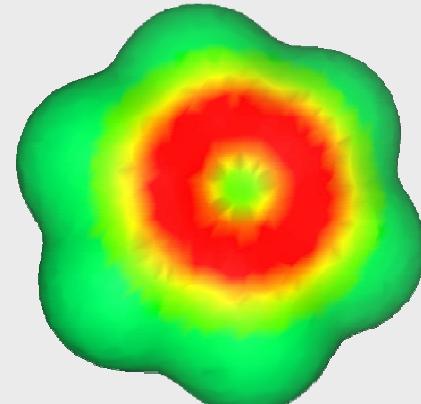
$$\log P_{\text{OW}}(\text{exp}) = 1.56 - 2.39$$

$$\log P_{\text{OW}}(\text{calc}) = 2.19$$

Local  
Ionization  
Energy  
(IEL)

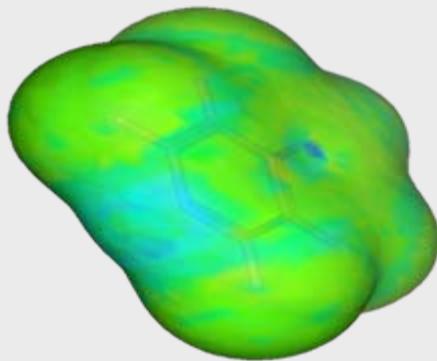


Local  
Electron  
Affinity  
(EAL)

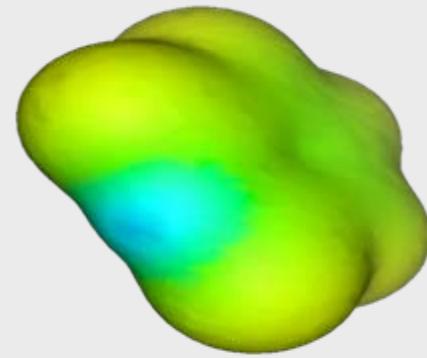


# Results: Surface Integral Models for $\log P_{\text{OW}}$ for Pyridine

Local  
Hydro-  
phobicity

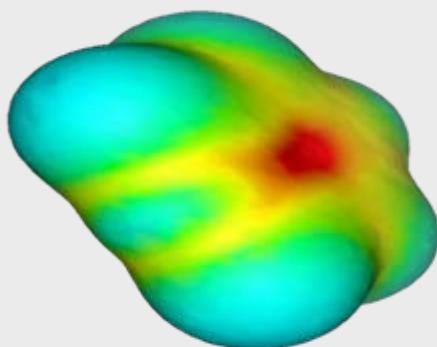


Local  
Molecular  
Electrostatic  
Potential  
(MEP)

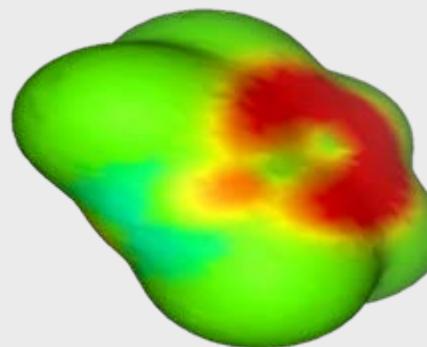


Pyridine  
 $\log P_{\text{OW}}(\text{exp}) = 0.44 - 1.09$   
 $\log P_{\text{OW}}(\text{calc}) = 0.98$

Local  
Ionization  
Energy  
(IEL)

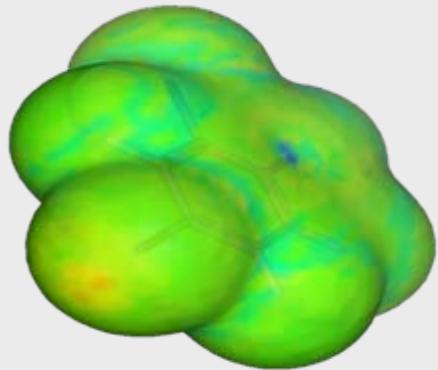


Local  
Electron  
Affinity  
(EAL)

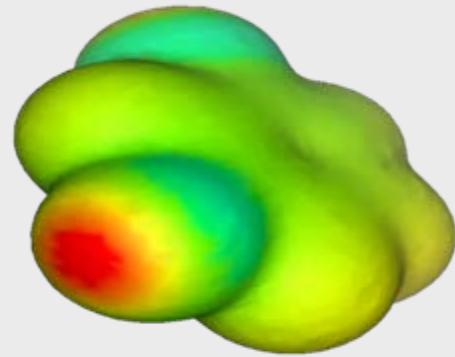


# Results: Surface Integral Models for $\log P_{\text{OW}}$ for 1,3 Dichlorobenzene

Local  
Hydro-  
phobicity

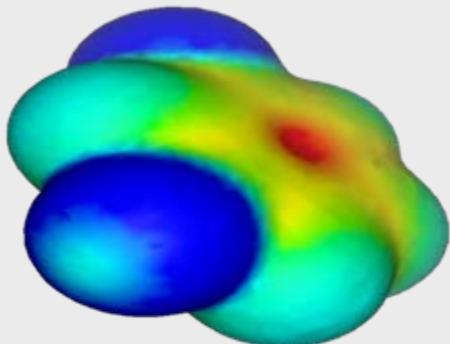


Local  
Molecular  
Electrostatic  
Potential  
(MEP)

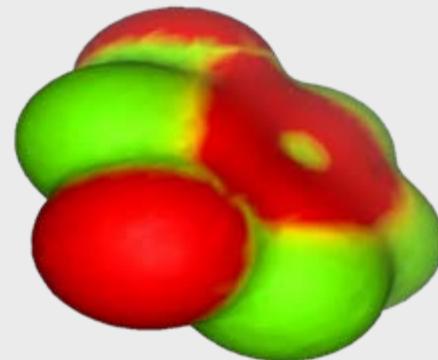


1,3 Dichlorobenzene  
 $\log P_{\text{OW}}(\text{exp}) = 3.38 - 3.62$   
 $\log P_{\text{OW}}(\text{calc}) = 3.41$

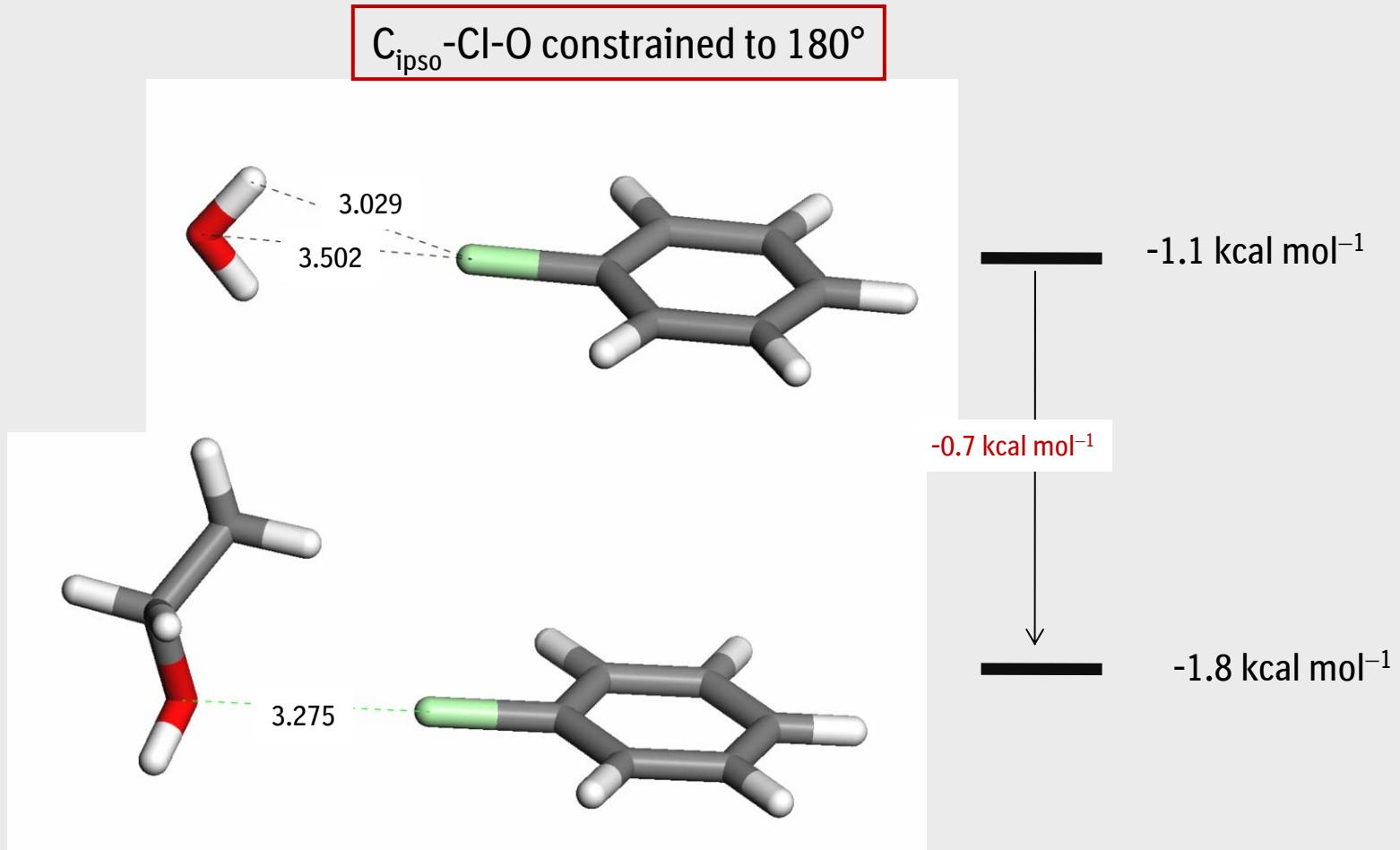
Local  
Ionization  
Energy  
(IEL)



Local  
Electron  
Affinity  
(EAL)



# MP2/aug-cc-PVTZ: Halogen Bond



# Summary and Outlook

- Local hydrophobicity defined;  $\text{Integral} = \log P_{\text{ow}}$
- Prediction performance comparable or better than state-of-the-art commercial models
- Prediction performance benefit from conformational sampling
- All calculations including QM  $\sim 10 - 20$ s for a single conformer
- Visualization
- Systematic improvements possible:
  - Choice of Hamiltonian (e.g. AM1\*, PM6)
  - Conformational Sampling
  - Use of Solvation model
- SIM models for different liquids / states:
  - Hexane/Water, Chloroform/Water, ...
  - Zwitterions, charged species ( $\rightarrow \log D_x$ )

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Thank you for your attention